# 66. Synthesis of Chiral Bicyclic Bis-lactam Components for the Controlled Self-Assembly of Hydrogen-Bonded Arrays 

by Marie-Josèphe Brienne, Jacqueline Gabard, Martine Leclercq, and Jean-Marie Lehn*<br>Chimie des Interactions Moléculaires, Collège de France, 11 place Marcelin Berthelot, F-75005 Paris<br>and Michel Chevé<br>Département de Chimie Pharmaceutique, Centre de Recherche Rhône-Poulenc Rorer, F-94403 Vitry sur Seine

(18.II.97)

The chiral biyclic bis-lactams of structures $\mathbf{3}$ and $\mathbf{4}$ were synthesized from the key intermediate $\mathbf{2}^{\mathbf{\prime}} \mathbf{b}$, the $N, N$-bis(4-methoxybenzyl) derivative of $2(\mathrm{X}=\mathrm{MeO})$ (Scheme 6 ). The synthesis of this intermediate involved two key steps: 1 ) a double condensation of glyoxylic acid/anisamide ( = oxoacetic acid/4-methoxybenzamide) adduct 11 c with veratrole ( 1,2 -dimethoxybenzene; $\mathbf{1 0}$ ) allowed the introduction of two glycine units at the 4,5 -positions of the veratrole ring to give 18c (Schemes 3 and 4);2) in order to circumvent the hydrolysis of 4 -methoxybenzoyl protective groups which proved to be unfeasible, these groups were transformed into 4-methoxybenzyl groups through a sequence involving thiocarbonylation followed by reduction (Scheme 5). Thereafter, the double intramolecular cyclization of the resulting diamino diester $22 \mathbf{c}$ proceeded easily to afford $\mathbf{2}^{\prime} \mathbf{b}$. This intermediate may be transformed via the tetrol $\mathbf{2}^{\prime} \mathrm{g}$ or the diol $\mathbf{2}^{\prime} \mathbf{h}$ into the $N$-protected derivatives of $\mathbf{2}$ ( $\mathrm{X}=\mathrm{OR}$ ) and of $\mathbf{3}$ ( $\mathrm{X}=\mathrm{OCOR}$ ). Cleavage of the 4 -alkoxybenzyl groups was achieved by ceric ammonium nitrate. However, when the aromatic ring bore ether functions ( $N$-protected 2 ), this normal reaction was accompanied by the oxidative ring cleavage to give the diene-diester structure 4 (Schemes 5 and 6 ).

Introduction. - In order to investigate the effect of molecular chirality on self-assembly, we previously studied the crystal structures of the H-bonded arrays formed by the racemate and by the enantiomerically pure ( - )-form of the bicyclic bis-lactam 1 [1]. In the racemate structure, the molecules are, as anticipated, assembled into an infinite undulating chain of alternating ( + )- and ( - )-units (see Fig., a). In the enantiomer structure, the molecules are assembled into cyclic tetrameric arrays different from the superstructure of racemic ( $\pm$ )-1. It was different from the hexameric assembly (see Fig. b)


1

$2 x=O R$
$3 \mathrm{x}=\mathrm{OCOR}$


4


Figure. Possible $H$-bonding patterns for a) ( $\pm$ )-1 and b) (-)-1
expected on the basis of internal H -bonding through the formation of cyclic bidentate interaction units as found in ( $\pm$ )-1. This structure is clearly not favorable from the point of view of the crystal packing, and we have not yet been able to find a guest molecule having the appropriate size and shape to fill up the cavity and enforce the crystallization in the expected way.

In the present paper, we describe another approach based on structural modifications of $\mathbf{1}$. The structures 2 or $\mathbf{3}$ seemed to us a good choice since they retained the unique H -bonding pattern of 2,5 -diazabicyclo[2.2.2]octane-3,6-dione (1) while allowing introduction of various substituents. Moreover, these modifications do not cause any stereochemical complications that would be encountered with the introduction of one or two substituents directly on the ethylene bridge. After several attempts to synthesize compounds 2, we succeeded in setting up a route which gave us access to $N$-protected compounds of type 2 (series $2^{\prime 1}$ )) and eventually to $N$-protected compounds of type 3 (series $3^{\prime}$ ). Unfortunately, for series $2^{\prime}$ the removal of the protective group could not be achieved without the cleavage of the aromatic ring to give compounds of structure 4. In series $\mathbf{3}^{\prime}$, however, the same deprotection easily gave the expected compounds 3 . These sets of synthetic investigations gave also rise to some novel reactions.

Synthetic Attempts towards Structures of Type 2. - Our initial efforts were concentrated on the synthesis of compounds $\mathbf{5}$ or $\mathbf{6}$, the direct precursors of structures 2 (Scheme 1).

Our first approach was similar to that used to synthesize 1 [1] (Scheme 2). The synthesis started with diester 7 obtained in five steps from 3,4-dimethoxybenzoic acid as described in [2]. Benzylic bromination of 7 was achieved by N -bromosuccinimide in $\mathrm{CCl}_{4}$ according to a described procedure [3] followed by treatment of the resulting mixture of diastereoisomers $\mathbf{8}$ by $\mathrm{NaN}_{3}$ in DMF to give 9 as a mixture of diastereoisomers. Attempts to transform 9 into the expected amino ester $\mathbf{5 b}$ were unsuccessful (catalytic hydrogena-

[^0]Scheme 1


6

Scheme 2

a) $N$-Bromosuccinimide, $\mathrm{CCl}_{4}$, $h \nu$, r.t., 16 h. b) $\mathrm{NaN}_{3}$, DMF, r.t., 4 h.
tion in the presence of $\mathrm{Pd} / \mathrm{C}$ or $\mathrm{PtO}_{2}$, Staudinger reaction [4]). Other efforts to obtain derivatives of amino ester $\mathbf{5 b}$ by substituting the Br -atoms of $\mathbf{8}$ by several nucleophiles (sodium phthalimide/DMF [5], $\mathrm{NaHNCOCF}_{3} / \mathrm{DMF}$ [6]) also failed.

Our next synthetic plan (Scheme 3) was centered around an intramolecular amidoalkylation (16a $\rightarrow$ 17a) using a methodology described in [7] [8]. Under acidic conditions [7], veratrole ( $=1,2$-dimethoxybenzene; 10) and glyoxylic acid/amide adducts $\mathbf{1 1}$ condensed easily to give the 3,4-dimethoxyphenylglycine derivatives 13. Hydrolysis of 13b ( 5 N HCl at reflux) followed by esterification of the resulting amino acid $14(\mathrm{HCl} / \mathrm{MeOH})$ afforded the amino ester 15, which was then condensed with the acid chloride 12a [8] to yield 16a. Surprisingly, while smooth intramolecular cyclization was reported [8] with the parent compound devoid of the MeO substituents in methanesulfonic acid, 16a only afforded complex reaction mixtures under the same acidic conditions. The failure of this intramolecular reaction is rather surprising since, as will be seen below, a similar intermolecular condensation ( $\mathbf{1 3} \boldsymbol{\rightarrow 1 8}$, Scheme 4), is possible.

Scheme 3

a) 11 (1 equiv.), $\mathrm{H}_{2} \mathrm{SO}_{4} / \mathrm{AcOH} 1: 9$, r.t., $1-2$ days. b) 5 N HCl , reflux, 18 h . c) $\mathrm{MeSO}_{3} \mathrm{H}$, MeOH , reflux, 18 h . d) $\mathbf{1 2 a}$ (1.1 equiv.), $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, r.t., 15 h .

An alternative route (Scheme 4) to compounds of type 5 was considered based on our discovery that, in methanesulfonic acid, 11b condensed with the acid 13b to give the $N, N^{\prime}$-dibenzoyl derivative $\mathbf{1 8 b}$ of $5\left(\mathrm{R}=\mathrm{Me}, \mathrm{R}^{\prime}=\mathrm{H}\right)$ as mixture of diastereoisomers $\left.\mathrm{A}^{2}\right)$ and $\mathbf{B}$ in $80 \%$ yield, while 11a and 13a gave only trace amounts of the corresponding $N, N^{\prime}$-bis(methoxycarbonyl) derivative. Other $N, N$-diacyl derivatives, $\mathbf{1 8 c}-\mathbf{f}$, were synthesized by the same procedure although in lower yields ( $45-55 \%$ ). In the two most favorable cases $\mathbf{1 8 b}$, $\mathbf{c}$, a one-pot synthesis was achieved from veratrole (10) and 11b, c ( 2 equiv.). The success of this reaction yet unreported may be attributed to the presence of the MeO groups in 3 and 4 positions which are known to facilitate Friedel-Crafts-type reactions. The acylamino acids $\mathbf{1 8 b}$, $\mathbf{c}$ were transformed into their dimethyl esters $\mathbf{1 9 b}$, $\mathbf{c}$ by usual acidic procedures ( $\mathrm{H}_{2} \mathrm{SO}_{4}$ in $\mathrm{MeOH}, \mathrm{H}_{2} \mathrm{SO}_{4}$ in $\mathrm{MeOH} / \mathrm{ClCH}_{2} \mathrm{CH}_{2} \mathrm{Cl}$ [9]). Esterification of $\mathbf{1 8 d}$ required neutral conditions such as diazomethane in $\mathrm{MeOH} / \mathrm{Et}_{2} \mathrm{O}$ (quant. yield).

Several attempts to determine the relative configuration meso or rac ( $\pm$ ) of these compounds by resolution of acids $\mathbf{1 8 b}$, $\mathbf{c}$ using chiral amines were unsuccessful. Finally, esters 19c $\mathbf{A}^{2}$ ) and 19cB were examined by HPLC on a chiral phase. While ester 19cA gave a single peak, ester $19 \mathbf{c B}$ was resolved into enantiomers thus establishing the relative configuration for 18cA and 19cA as meso and for 18 cB and 19 cB as rac. Comparison of NMR spectra of the series 18 and 19 (Table 4 in Exper. Part) suggests, in particular, the same relative configurations ( $\mathbf{A}=$ meso and $\mathbf{B}=r a c$ ) for diastereoisomers $\mathbf{1 8 b}$ and $\mathbf{1 9 b}$. For all other compounds, however, no conclusion can be drawn. It should be noted that

[^1]
a) 11 (1 equiv.), $\mathrm{MeSO}_{3} \mathrm{H}$, r.t., 2 days. b) Method $A: \mathrm{H}_{2} \mathrm{SO}_{4}, \mathrm{MeOH}$, reflux, 17 h ; Method $B: \mathrm{H}_{2} \mathrm{SO}_{4}$, MeOH , 1,2-dichloroethane, reflux, 24 h ; Method $C: \mathrm{CH}_{2} \mathrm{~N}_{2}, \mathrm{Et}_{2} \mathrm{O} / \mathrm{MeOH}$, r.t. $c$ ) 19d, $N$-piperidinothiourea, AcOH, abs. EtOH, reflux, $4 \mathrm{~h} . d) \mathrm{MeOH}$, reflux, 1.5 h .
in most of the cases, in the reaction giving rise to compounds 18, isomer $\mathbf{A}$ (proved to be meso in series $\mathbf{b}$ and $\mathbf{c}$ ) was always the major compound.

All attempts to hydrolyze the $N$-acyl groups of $\mathbf{1 8 b} \mathbf{- d}$ or $\mathbf{1 9 b}-\mathbf{d}$ under either acidic or basic standard procedures were unsuccessful. However, we were able to cleave the $N$-(chloroacetyl) groups of 19d by means of $N$-piperidinothiourea [10] producing in fair yields ( $40-45 \%$ ) the amino-tetrahydro-oxoisoquinolinecarboxylate 6a resulting from the intramolecular cyclization of the intermediate amino ester $\mathbf{5 a}$ (not isolated). Attempts to achieve the second intramolecular cyclization to give 2a were unsuccessful ( MeONa in $\mathrm{MeOH}, \mathrm{NaCN}$ in MeOH ) due to formation of complex product mixtures. Even in the absence of an alkaline catalyst, the tetrahydroisoquinolinecarboxylate 6 a in refluxing MeOH gave very rapidly a nearly quantitative formation of a yellow compound. Based on NMR data and elemental analysis, the dihydroisoquinolininecarboxylate structure 20, resulting of the loss of $\mathrm{NH}_{3}$, was attributed to this compound. By action of a base, 20 was further decomposed to give complex mixtures.

Synthesis of $\mathbf{N}$-Protected 2 (Series $\mathbf{2}^{\prime}$ ) and Formation of Cleavage Compounds of Type 4. - The easy accessibility of the bis(acylamino) derivatives $19 \mathrm{~b}, \mathrm{c}$ encouraged us to use these compounds to reach our target structure 2 . Since the cleavage of amide groups proved to be unfeasible, we considered that the transformation of the benzoylamino into benzylamino groups would be a good alternative.

Scheme 5


24'


4'a $Y=\mathrm{PhCH}_{2}$


2a
4b $Y=H$
a) Lawesson's reagent, THF, $60^{\circ}, 18-24 \mathrm{~h}$. b) Raney-Ni, $\mathrm{MeOH} / \mathrm{THF}$, r.t., $8 \mathrm{~min} . c$ ) $\mathrm{CHCl}_{3}$, r.t., 2 days.
d) $\mathrm{NaOMe}, \mathrm{MeOH}$, r.t., 24 h. e) $\mathrm{CAN}, \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}$, r.t., $1.5 \mathrm{~h} . f$ ) $\mathrm{I}_{2}$, toluene, $h v$, reflux, 2.5 h .

Among the methods available for the reduction of amides to amines we chose a procedure already described [11] via the corresponding thioamides, compatible with the presence of ester groups (Scheme 5). Amides 19b, c were readily transformed to bis(thiobenzoyl) derivatives 21b, c by using Lawesson's reagent in THF [12] (70\% yield). Subsequent treatment with Raney- Ni (in THF/MeOH) afforded the corresponding dibenzyl derivatives 22b, $\mathbf{c}$. The reduction was very fast (in $c a .8 \mathrm{~min}$ ). The reaction was stopped when the initial yellow color of the solution disappeared. Longer reaction times led to recoloration of the reaction mixture and formation of undesired products. These amino esters 22b, c cyclized spontaneously to give amino-tetrahydro-oxoisoquinolinecarboxylates 23b, c. Since no epimerization seemed to occur during cyclization, the meso-isomer of $\mathbf{2 1 b}, \mathbf{c}$ gave cis-23b, c while the rac-isomer of 21b, $\mathbf{c}$ gave trans-23b, $\mathbf{c}$. In contrast to the results observed for $\mathbf{6 a}$ (see above), intramolecular lactamization of 23b and 23c took place readily under the same conditions ( $\mathrm{NaOMe} / \mathrm{MeOH}$ ) to give $\mathbf{2}^{\prime}$ a and $\mathbf{2}^{\prime} \mathbf{b}{ }^{1}$ ), respectively, in $50-60 \%$ overall yield from 21b and 21c. Moreover, lactamization proceeded on both diastereoisomers as well, meaning that epimerization of trans-into cis-23 took place before lactamization.

Debenzylation of $\mathbf{2}^{\prime}$ a using several known methods [13] was unsuccessful. Among the methods examined were: $\mathrm{H}_{2}, \mathrm{Pd} / \mathrm{C}, \mathrm{AcOH} ; \mathrm{Pd} / \mathrm{C}, \mathrm{HCO}_{2} \mathrm{NH}_{4}, \mathrm{MeOH}$, reflux [14]; $t$-BuLi, THF then $\mathrm{O}_{2}$ [15]; $t$-BuOK, DMSO then $\mathrm{O}_{2}$ [16]; $\mathrm{Li}, \mathrm{NH}_{3}$ [17]; $\mathrm{BBr}_{3}$, xylene or mesitylene, reflux [18]. Under these last conditions, demethylation was the unique reaction observed. In all other cases, either starting material was recovered or complex reaction mixtures due to degradation of the bicyclic system were obtained.

Cerium ammonium nitrate $\left(\mathrm{Ce}\left(\mathrm{NH}_{4}\right)_{2}\left(\mathrm{NO}_{3}\right)_{6}\right.$; CAN) is an effective reagent for the removal of a 4-methoxybenzyl group on a N -atom [19]. Under these conditions the $N$-benzyl groups are not cleaved. We previously used this method with success $(85 \%$ yield) for deblocking the $N, N^{\prime}$-bis(4-methoxybenzyl) derivative of $1\left(1^{\prime}\right)$ [1]. Treatment of $\mathbf{2}^{\prime} \mathbf{b}$ under the same conditions also cleaved the protective groups but the only isolated product ( $30-40 \%$ yield) was not the desired product 2a. From elemental analysis, the empirical formula of that new compound was $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{6}$ (280.24) instead of $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{4}$ (248.24) for the expected compound $\mathbf{2 a}$. Mass spectrometry was at first a bit confusing since besides a large signal $[M+1]^{+}$at 281.3 , a signal at 249.3 , almost as intense, was also present. Moreover, the presence in the IR spectrum of a $\mathrm{C}=\mathrm{O}$ band at $1730 \mathrm{~cm}^{-1}$ clearly indicated the presence of an ester group. All these data are consistent with the structure of $(E, E)$-diene-diester $\mathbf{4 b}$. This means that besides the normal cleavage of the $N$-(4-methoxybenzyl) group of $\mathbf{2}^{\prime} \mathbf{b}$ under the action of CAN, the aromatic ring was oxidized simultaneously. Since CAN does not affect the benzyl groups, the same treatment converted $\mathbf{2}^{\prime} \mathbf{a}$ to the dibenzyl derivative $\mathbf{4}^{\prime} \mathbf{a}$. The ( $E, E$ ) configuration of the diene-diester $4^{\prime}$ a was confirmed by its isomerization into the ( $Z, Z$ )-derivative $\mathbf{2 4}^{\prime}$ in the presence of $I_{2}$. Moreover, 4'a was easily hydrogenated to give a mixture of isomers which were not studied further. NMR Data of these compounds discussed below confirmed their structures.

To our knowledge, such an oxidative cleavage of an o-dimethoxyaryl ring by CAN has not yet been reported and would deserve a more thorough study.

Synthesis of Compounds of Types 3 and 4. - The $N$-(4-methoxybenzyl) derivative 2'b turned out to be a good intermediate to the synthesis of various diesters of type 4 and eventually of type 3 (Scheme 6).

Treatment of $\mathbf{2}^{\prime} \mathbf{b}$ with $\mathrm{BBr}_{3}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ afforded tetrol $\mathbf{2}^{\prime} \mathbf{g}$ in good yield. Realkylation by various alkyl bromides to give compounds $\mathbf{2}^{\prime} \mathbf{b}-\mathbf{e}$, followed by treatment with CAN yielded diene-diesters $\mathbf{4 b}, \mathbf{c}, \mathbf{e}$. The reaction with CAN proceeded in the same manner as for the parent compound $\mathbf{2}^{\prime} \mathbf{b}$ (see above), although a significant decrease of the reaction yield was observed with the lengthening of the ether chains (only $10 \%$ for $\mathbf{4 e}$ ) due, in particular, to a decreasing solubility.

The $N$-(4-hydroxybenzyl) groups of tetrol $\mathbf{2}^{\prime} \mathbf{g}$ could be selectively methylated in the presence of borax known to protect the ortho-dihydroxy functions [20] to give $\mathbf{2}^{\prime} \mathbf{h}$ in good yield ( $>80 \%$ ). This diol was realkylated by dodecyl bromide to give $\mathbf{2}^{\prime} \mathbf{f}$ or esterified with various acyl chlorides to yield compounds $\mathbf{3} \mathbf{3} \mathbf{a}-\mathbf{e}$. Treatment of the latter with CAN afforded esters $\mathbf{3 a}-\mathbf{e}$. It is noteworthy that the aromatic ring bearing ester functions (series $\mathbf{3}^{\prime}$ ) remains unaffected by the oxidative treatment with CAN, in contrast to the cleavage observed when the aromatic ring bears ether groups (series $\mathbf{2}^{\prime}$ ).

Scheme 6


a) $\mathrm{BBr}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, r.t., 5 h. b) $\mathrm{K}_{2} \mathrm{CO}_{3}$, alkyl bromide, DMF, $80^{\circ}, 15 \mathrm{~h}$. c) $\mathrm{CAN}, \mathrm{MeCN} / \mathrm{H} 2 \mathrm{O}$, r.t., $1.5-3 \mathrm{~h}$. d) 1) NaOH , borax, $\mathrm{H}_{2} \mathrm{O}, 1 \mathrm{~h}$; 2) $\mathrm{Me}_{2} \mathrm{SO}_{4}, \mathrm{NaOH}, 3 \mathrm{~h}$; 3) $3.6 \mathrm{~N} \mathrm{H}_{2} \mathrm{SO}_{4}$, reflux, 15 min .

Resolution of 3c. - We previously described the easy resolution of the $N, N^{\prime}$-bis(4-methoxybenzyl) derivative of $\mathbf{1}\left(1^{\prime}\right)$ [1] on the gram scale by HPLC on tris( 3,5 -dimethylphenylcarbamoyl)cellulose (Chiralcel OD). The same type of phases, Chiralcel OD and OC (tris(phenylcarbamoyl)cellulose), was less efficient for compounds $\mathbf{2}^{\prime}$ a and $\mathbf{3}^{\prime} \mathbf{c}$ and gave satisfactory results only on the analytical scale. In contrast, compound 3 c was nicely resolved on Chiralcel $O C$. In view of this result, we considered the possibility of transforming 3 c into the diol $2(\mathrm{X}=\mathrm{OH})$ which eventually would give access to a variety of compounds 2 or 3 in racemic or enantiomeric forms. Unfortunately, all attempts to hydrolyze the ester functions of 3 c under either acidic or basic standard procedures were unsuccessful.

NMR Spectral Features of Compounds of Types 2-4. - All assignments of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ resonances (Tables 1 and 2), and ${ }^{13} \mathrm{C}$-NMR resonances (Table 3) were obtained by conventional decoupling irradiations, NOESY, DEPT, or INEPT techniques. All spectra were perfectly consistent with the $C_{2}$ symmetry of the molecules.

Table 1. Selected ${ }^{2} H-N M R$ Chemical Shifts [ppm] for Compounds $\mathbf{2}^{\prime}-\mathbf{4}^{\prime}, \mathbf{3}, \mathbf{4}$, and $\mathbf{2 4}^{\prime}$ in $\mathrm{CDCl}_{3}{ }^{\text {a }}$ ). For convenience, all atoms are numbered as for 2 (arbitrary numbering).

|  | 2'a | 2'b | 2'c | 2'd | $2^{\prime} \mathrm{e}^{\text {b }}$ ) | 2'f |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{H}-\mathrm{C}(3)^{\mathrm{c}}$ ) | 4.68 | 4.66 | 4.63 | 4.63 | 4.78 | 4.64 |  |
| $\mathrm{H}-\mathrm{C}(9)^{\text {c }}$ ) | 6.50 | 6.49 | 6.52 | 6.52 | 6.84 | 6.52 |  |
| $\mathrm{CH}_{A} \mathrm{H}_{B}-\mathrm{N}^{\text {d }}$ ) | 4.79-4.36 | 4.75-4.30 | 4.71-4.30 | $4.70-4.30$ | 4.62-4.46 | 4.70-4.32 |  |
|  | 3'a | 3'b | 3'c | 3'd | 3 e | $4^{\prime} \mathbf{a}$ | 24 |
| $\mathrm{H}-\mathrm{C}(3)^{\text {c }}$ ) | 4.79 | 4.72 | 4.72 | 4.71 | 4.71 | 4.37 | 6.14 |
| $\mathrm{H}-\mathrm{C}(9)^{\mathrm{c}}$ ) | 7.07 | 6.88 | 6.86 | 6.86 | 6.86 | 6.05 | 6.18 |
| $\mathrm{CH}_{A} \mathrm{H}_{B}-\mathrm{N}^{\mathrm{d}}$ ) | 4.70-4.42 | 4.61-4.40 | 4.63-4.39 | 4.62-4.47 | 4.61-4.40 | 4.73-4.48 | 4.76-4.40 |
|  | 3d | 3 e | 4c | 4e |  |  |  |
| $\mathrm{H}-\mathrm{C}(3)^{\mathrm{e}}$ ) | 4.79 | 4.79 | 4.45 | 4.44 |  |  |  |
| $\mathrm{H}-\mathrm{C}(9)^{\text {c }}$ ) | 7.28 | 7.24 | 6.34 | 6.33 |  |  |  |
| $\mathrm{NH}^{\mathrm{e}}$ ) | 7.28 | $7.11^{\mathrm{f}}$ ) | 7.29 | 6.98 |  |  |  |

${ }^{\text {a }}$ ) Resonances for other protons, see Exper. Part.
${ }^{\text {b }}$ ) $\operatorname{In}\left(\mathrm{D}_{8}\right)$ THF.
${ }^{\text {c }} \mathrm{s}$.
$\left.{ }^{\text {d }}\right) \quad J(A, B)[\mathrm{Hz}]: 14.6\left(\mathbf{2 4}^{\prime}\right) ; 14.7\left(\mathbf{2}^{\prime} \mathbf{b}, \mathbf{e}, \mathbf{3}^{\prime} \mathbf{a} \mathbf{a} \mathbf{e}\right) ; 14.8\left(\mathbf{2}^{\prime} \mathbf{c}, \mathbf{d}, \mathbf{f}\right) ; 14.9\left(\mathbf{2}^{\prime} \mathbf{a}\right) ; 15.1$ (4'a).
$\left.{ }^{e}\right) \quad m, A A^{\prime} X X^{\prime}$ spin system.
$\left.{ }^{f}\right) \quad$ Broad $s$.

Table 2. Selected ${ }^{1} H-N M R$ Chemical Shifts $[\mathrm{ppm}]$ for Compounds $\mathbf{2}^{\prime}, \mathbf{4}^{\prime}, \mathbf{3}$, and 4, in $\left(D_{6}\right)$ DMSO $\left.^{\mathrm{a}}\right)$.
For convenience, all atoms are numbered as for 2 (arbitrary numbering).

|  | $\mathbf{2}^{\prime} \mathbf{a}$ | 2'b | 2'g | 2'h | 4'a |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{H}-\mathrm{C}(3)^{\text {b }}$ ) | 5.09 | 5.02 | 4.88 | 4.94 | 5.04 |  |
| $\mathrm{H}-\mathrm{C}(9)^{\mathrm{b}}$ ) | 7.09 | 7.03 | 6.82 | 6.83 | 6.46 |  |
| $\mathrm{CH}_{A} \mathrm{H}_{B}-\mathrm{N}^{\mathrm{c}}$ ) | 4.58-4.32 | 4.48-4.25 | 4.42-4.09 | 4.47-4.17 | 4.64-4.36 |  |
|  | 3a | 3b | 3c | 3d | 3 e | 4b |
| $\mathrm{H}-\mathrm{C}(3)^{\text {d }}$ ) | 4.94 | 4.88 | 4.87 | 4.88 | 4.86 | 4.59 |
| $\mathrm{H}-\mathrm{C}(9)^{\text {b }}$ ) | 7.80 | 7.53 | 7.53 | 7.53 | 7.52 | 6.52 |
| $\mathrm{NH}^{\text {d }}$ ) | 9.34 | 9.27 | 9.26 | 9.26 | 9.26 | 9.13 |

[^2]Table 3. Selected ${ }^{13} \mathrm{C}$-NMR Chemical Shifts [ppm] for Compounds $\mathbf{2 '}^{\prime} \mathbf{a}, \mathbf{2} \mathbf{\prime} \mathbf{b}, \mathbf{4}^{\prime} \mathbf{a}, \mathbf{3 a}, \mathbf{4 b}$; and $\mathbf{2 4}^{\prime}$, in $\mathrm{CDCl}_{3}$ and $\left(D_{6}\right) D M S O^{2}$ ). For convenience, all atoms are numbered as for $\mathbf{2}$ (arbitrary numbering).

|  | $\mathrm{CDCl}_{3}$ |  |  | ( $\mathrm{D}_{6}$ ) DMSO |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 2'a | 4'a | $24^{\prime}$ | 2'a | 2'b | 3a | 4'a | 4b |
| C(2) | 168.8 | 166.0 | 165.6 | 168.7 | 168.6 | 171.3 | 166.1 | 169.2 |
| C(3) | 64.2 | 66.9 | 59.4 | 63.9 | 63.8 | 58.9 | 66.0 | 61.3 |
| C(8) | 129.9 | 138.5 | 144.1 | 130.4 | 130.4 | 137.9 | 138.9 | 140.5 |
| C(9) | 108.0 | 123.8 | 115.3 | 109.3 | 109.2 | 120.1 | 123.6 | 123.1 |
| C(10) | 148.2 | 164.5 | 164.3 | 147.9 | 147.9 | 140.9 | 164.6 | 165.0 |

${ }^{\text {a }}$ ) Resonances for other C-atoms, see Exper. Part.

For all $N$-protected compounds (prime series), $\mathrm{H}-\mathrm{C}(3)$ appeared as a $s$ in the ${ }^{1} \mathrm{H}-\mathrm{NMR}$ while for deprotected compounds, $\mathrm{H}-\mathrm{C}(3)$ formed with NH a $A A^{\prime} X X^{\prime}$ spin system and appeared as a $m$. For all compounds of the series $\mathbf{2}^{\prime}, \mathbf{3}$, and $\mathbf{3}^{\prime}$, the $\mathrm{H}-\mathrm{C}(3)$ signal was found within the range $4.6-4.8 \mathrm{ppm}$ in $\mathrm{CDCl}_{3}$ (Table 1) and $4.9-5.1 \mathrm{ppm}$ in ( $\mathrm{D}_{6}$ )DMSO (Table 2). Considering that the bridgehead proton of 1 resonated at 3.68 ppm in $\left(\mathrm{D}_{6}\right) \mathrm{DMSO}$ and
 $\Delta \delta$ of the aromatic fused ring on $\mathrm{H}-\mathrm{C}(3)$ was $0.6-0.8$ in $\mathrm{CDCl}_{3}$ and $0.9-1.2$ in ( $\mathrm{D}_{5}$ ) DMSO, values only slightly lower than those observed ( $\Delta \delta 1.4$ ) in the bicyclo[2.2.2]octane series [21]. In the exocyclic diene structures such as $\boldsymbol{4}^{\prime} \mathbf{a}$ and $\mathbf{4 b}, \mathrm{c}, \mathrm{e}$, the $\mathrm{H}-\mathrm{C}(3)$ signal was shifted to low fields by 0.4 ppm in $\mathrm{CDCl}_{3}$ and by $c a .1 \mathrm{ppm}$ in ( $\mathrm{D}_{6}$ )DMSO. This effect may be compared to the effect observed for some vicinal dimethylidene derivatives of bicyclo[2.2.2]octane [22] ( $4 \delta 0.4-0.7$ ), although in series 4 and $4^{\prime}$ the $\mathrm{C}=\mathrm{C}$ bonds are certainly not coplanar due to a severe steric hindrance between the $\mathrm{CO}_{2} \mathrm{Me}$ groups. By contrast, for $24^{\prime}$ with the ( $Z, Z$ )-configuration, the $\mathrm{H}-\mathrm{C}(3)$ signal was shifted downfield by $1.8 \mathrm{ppm}(\delta 6.14)$ as compared with its ( $E, E$ )-isomer $4^{\prime}$ a. Obviously, this shift was caused by the anisotropic effect of the ester carbonyl group and confirmed the configurational assignments. Another evidence for these assignments was the observation of a NOE at $\mathbf{H}-\mathrm{C}(9)$ on irradiation of $\mathrm{H}-\mathrm{C}(3)$ in $\mathbf{4}^{\prime} \mathrm{a}$ and $\mathbf{2}^{\prime}$ a indicating the proximity of these protons.

For all compounds of the series $\mathbf{2}^{\prime}$ and $\mathbf{3}^{\prime}, \delta$ values in $\mathrm{CDCl}_{3}$ for $\mathbf{H}-\mathrm{C}(9)\left(s\right.$ at $c a .6 .5$ for $\mathbf{2}^{\prime}$ and 6.9 for $\mathbf{3}^{\prime}$, Table 1) were in good agreement with the predicted values based on empirical increments for aromatic substituents [23]. As expected, this proton in the diene series 4 and $4^{\prime}$ appeared at higher field. On the other hand, the inside proton of $24^{\prime}$ was slightly deshielded ( $\delta 6.18$ ) as compared to the outside proton of its isomer $\mathbf{4}^{\prime} \mathbf{a}(\delta 6.05)$. This deshielding may result from the anisotropic effect of the $\mathrm{C}=\mathrm{C}$ bond as it was observed in dimethylidene derivatives of bicyclo[2.2.2]octane [22].

Conclusion. - The present synthetic work gives access to the highly functionalized bicyclo[2.2.2]octane structures $\mathbf{3}$ and 4. It revealed novel reactions that could possess further synthetic potential. Furthermore, in view of the H -bonding features of the parent compound $\mathbf{1}$ [1], the derivatives of types $\mathbf{3}$ and $\mathbf{4}$ may also form superstructures displaying the effect of molecular chirality on self-assembly [24]. In addition, such arrays may present novel physicochemical properties, as was the case in the generation of liquid crystalline phases through the self-assembly of complementary H -bonded components [25] [26].

We thank J. Jacques for helpful and stimulating discussions, N. Pustet and A. Mannschreck, Universität Regensburg, for performing some anal. HPLC resolutions, and L. Lacombe for determining the NMR spectra.

## Experimental Part

General. The following reagents were commercially available: $\alpha$-hydroxyhippuric acid (= (benzoylamino)hydroxyacetic acid; 11b, Aldrich), glyoxylic acid monohydrate ( $=$ oxoacetic acid; Aldrich or Janssen Chimica),

Lawesson's reagent (Aldrich), methanesulfonic acid (puriss.; Fluka). Raney-Ni (Aldrich or Prolabo), ceric ammonium nitrate (CAN; Aldrich), substituted benzamides and chloroacetamide (Aldrich). TLC: Merck silica gel 60 F254, 0.25 mm (anal. TLC), 1 or 2 mm (prep. TLC), UV detection. Prep. column chromatography (CC): Merck silica gel $60(0.040-0.063 \mathrm{~mm})$ or $\mathrm{Al}_{2} \mathrm{O}_{3} 90$ activity II-III. M.p. and enthalpies of fusion ( 4 H ): Perkin-Elmer$D S C 7$ differential scanning calorimeter, heating rate $5^{\circ} / \mathrm{min}$ unless otherwise state, temperatures in ${ }^{\circ}, \Delta \mathrm{H}$ in $\mathrm{kJ} / \mathrm{mol}$. IR: Perkin-Elmer 297, KBr pellets. NMR: Bruker AM200SY equipped with a data system Aspect 3000, $\delta$ in ppm downfield from $\mathrm{SiMe}_{4}, J$ in Hz . MS: FAB (positive). Elemental analyses: ‘Service Central d’Analyse du CNRS'.

Diethyla, $\alpha^{\prime}$-Dibromo-4,5-dimethoxybenzene-1,2-diacetate (8). A stirred mixture of $7(2.4 \mathrm{~g}, 7.74 \mathrm{mmol})$, prepared according to [2], $N$-bromosuccinimide (NBS; $2.76 \mathrm{~g}, 15.86 \mathrm{mmol}$ ) and $\mathrm{CCl}_{4}(30 \mathrm{ml})$ was irradiated with two $100-\mathrm{W}$ lamps for 16 h . The insoluble $N$-succinimide was filtered off and the filtrate evaporated to give $8(3.6 \mathrm{~g})$ as a mixture of 2 diastereoisomers which was used without further purification.

Diethyl $\alpha, \alpha^{\prime}$-Diazido-4,5-dimethoxybenzene-1,2-diacetate (9). A mixture of crude 8 ( $1.9 \mathrm{~g}, 4 \mathrm{mmol}$ ), $\mathrm{NaN}_{3}$ $(0.57 \mathrm{~g}, 8.8 \mathrm{mmol})$, and DMF ( 5 ml ) was stirred at r.t. for 4 h , then diluted with $\mathrm{H}_{2} \mathrm{O}$, and extracted with $\mathrm{Et}_{2} \mathrm{O}$. After evaporation, the residue was purified by $\mathrm{CC}\left(\mathrm{SiO}_{2}, 50 \mathrm{~g}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to give 9 as a $1: 1$ mixture of 2 diastereoisomers. IR (film): 2100s, 1740s, $1610 \mathrm{~m} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 6.91,6.88(2 \mathrm{~s}$, arom. H); 5.42, 5.38 ( 2 s , $\mathrm{CH}) ; 4.25,4.30\left(2 q, \mathrm{CH}_{2}\right) ; 3.9(s, \mathrm{MeO}) ; 1.26,1.25(2 t, \mathrm{Me})$.

Hydroxy/(methoxycarbonyl)amino Jacetic Acid (11a). Prepared according to [8].
Hydroxy/(4-methoxybenzoyl)amino lacetic Acid (11c). A mixture of glyoxylic acid monohydrate ( 18.4 g , 0.2 mol ) and anisamide ( $30.25 \mathrm{~g}, 0.2 \mathrm{~mol}$ ) in acetone ( 200 ml ) was refluxed with stirring. Dissolution occurred after a few minutes followed by the progressive apparition of a precipitate. After 1 h , more glyoxylic acid monohydrate $(9.2 \mathrm{~g}, 0.1 \mathrm{~mol})$ and acetone ( 200 ml ) were added, and reflux was continued for 23 h . The mixture was cooled to r.t. and left overnight. The solid 11 c was collected by filtration ( $33 \mathrm{~g}, 73 \%$ ). M.p. $148.5^{\circ}$ (dec.; $10^{\circ} / \mathrm{min}$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ $\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right): 9.09(d, J=8, \mathrm{NH}) ; 7.87,6.98$ (arom. H); $5.57(d, J=8, \mathrm{CH}) ; 3.80(s, \mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{NO}_{5}$ (225.20): C 53.33, H 4.92, N 6.22; found: C 53.4, H 4.9, N 6.2 .
[(Chloroacetyl)amino]hydroxyacetic Acid (11d). A mixture of glyoxylic acid monohydrate ( $9.2 \mathrm{~g}, 0.1 \mathrm{~mol}$ ) and chloroacetamide ( $9.4 \mathrm{~g}, 0.1 \mathrm{~mol}$ ) was heated on a rotatory evaporator at $60^{\circ}$ for 30 min under 1 atm pressure and then for 1 h in vacuo. The gummy residue was dissolved in $\mathrm{AcOEt}(50 \mathrm{ml})$ at $60^{\circ}$ and the soln. cooled to r.t. and allowed to crystallize overnight. The crystalline precipitate was collected by filtration ( $7.85 \mathrm{~g}, 47 \%$ ). M.p. ca. $120^{\circ} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right): 8.97(d, J=8, \mathrm{NH}) ; 5.38(d, J=8, \mathrm{CH}) ; 4.10\left(s, \mathrm{CH}_{2}\right)$. ([27]: M.p. $105^{\circ}{ }^{1} \mathrm{H}-\mathrm{NMR}$ : same $\delta$ except for $\mathrm{NH}, \delta 8.8$ ). Anal. calc. for $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{ClNO}_{4}$ (167.55): C $28.67, \mathrm{H} 3.61, \mathrm{Cl} 21.16, \mathrm{~N} 8.36$; found: C 28.75, H 3.7, Cl 20.9, N 8.4.

Hydroxy/(4-nitrobenzoyl)amino]acetic Acid (11e). An intimate mixture of solid glyoxylic acid monohydrate $(5.55 \mathrm{~g}, 60 \mathrm{mmol})$ and 4-nitrobenzamide $(10.05 \mathrm{~g}, 60 \mathrm{mmol})$ was heated on a rotatory evaporator at $60^{\circ}$, for 20 min at 1 atm and then for 20 min in vacuo. Acetone ( 500 ml ) was added, and the suspension refluxed for 3 h with stirring and then evaporated. The glassy residue containing ca. $70 \%$ of $11 \mathbf{e}$ and $30 \%$ of 4 -nitrobenzamide was used without further purification. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right): 9.63(d, J=8, \mathrm{NH}) ; 8.30,8.10(4 \operatorname{arom} . \mathrm{H}) ; 5.60(d, J=8$, CH ).

Hydroxy[(2-nitrobenzoyl)amino Jacetic Acid (11f). As described for 11e, from glyoxylic acid monohydrate $(2.76 \mathrm{~g}, 30 \mathrm{mmol})$ and 2 -nitrobenzamide ( $5 \mathrm{~g}, 30 \mathrm{mmol}$ ). The residue containing ca. $70 \% 11 \mathrm{f}$ and $30 \%$ of 2 -nitrobenzamide was used without further purification. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right): 9.52(d, J=8.3, \mathrm{NH}) ; 8.00,7.7$ ( 4 arom. H); 5.55 ( $d, J=8.3, \mathrm{CH}$ ).

Acids 13a-f: General Procedure. The procedure described in [7] was slightly modified as following. To a cooled and stirred soln. (ca. $10^{\circ}$ ) of veratrole ( $\mathbf{1 0} ; 0.2 \mathrm{~mol}$ ) in conc. $\mathrm{H}_{2} \mathrm{SO}_{4}(10 \mathrm{ml})$ and $\mathrm{AcOH}(90 \mathrm{ml})$ was added the appropriate reagent $11 \mathrm{a}-\mathrm{f}(0.1 \mathrm{~mol})$. Stirring was continued for $24-48 \mathrm{~h}$ at r.t. The mixture was poured into ice and $\mathrm{H}_{2} \mathrm{O}$ and the resulting precipitate collected by filtration, washed successively with $\mathrm{H}_{2} \mathrm{O}$ and pentane, and then recrystallized.

3,4-Dimethoxy- $\alpha$-[(methoxycarbonyl)amino]benzeneacetic Acid (13a): Recrystallized from AcOEt/hexane. Yield $81 \%$. M.p. $138^{\circ} . \Delta H=30.7 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right): 7.87(d, J=8, \mathrm{NH}) ; 6.98$ ( $s, 1$ arom. H ); 6.89 $(d, 2$ arom. H$) ; 5.02(d, J=8, \mathrm{CH}) ; 3.72,3.71(2 s, 2 \mathrm{MeO}) ; 3.54(s, \mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{NO}_{6}(269.25)$ : C 53.53 , H $5.62, \mathrm{~N} 5.20$; found: C $53.6, \mathrm{H} 5.5$, N 5.0.
$\alpha$-(Benzoylamino)-3,4-dimethoxybenzeneacetic Acid (13b): Recrystallized from methoxyethanol/ $\mathrm{H}_{2} \mathrm{O}$. Yield $88 \%$. M.p. $211^{\circ} . \Delta H=53.4 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right): 8.93(d, J=7, \mathrm{NH}) ; 7.88,7.48$ ( $2 m, \mathrm{PhCO}$ ); 7.11 ( $d, 1$ arom. H); $6.97(m, 2$ arom. H$) ; 5.49(d, J=7, \mathrm{CH}) ; 3.75,3.74(2 s, 2 \mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{17} \mathrm{H}_{17} \mathrm{NO}_{5}$ (315.33): C $64.75, \mathrm{H} 5.43$, N 4.44 ; found: C $64.7, \mathrm{H} 5.4$, N 4.3.

3,4-Dimethoxy- $\alpha$-[(4-methoxybenzoyl)amino]benzeneacetic Acid (13c): Recrystallized from methoxyethanol/ $\mathrm{H}_{2} \mathrm{O}$. Yield $73 \%$. M.p. 179 and $198^{\circ} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right): 8.74(d, J=7, \mathrm{NH}) ; 7.88,6.97$ ( $2 d, \mathrm{MeOC}_{6} \mathrm{H}_{4} \mathrm{CO}$ ); $7.09(d, 1$ arom. H); $6.97(m, 2$ arom. H$) ; 5.47(d, J=7, \mathrm{CH}) ; 3.79,3.74,3.72(3 \mathrm{~s}, 3 \mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{NO}_{6}$ (345.36): C 62.60, H 5.55, N 4.06; found: C $62.5, \mathrm{H} 5.6, \mathrm{~N} 4.0$.
$\alpha$-/(Chloroacetyl)amino ]-3,4-dimethoxybenzeneacetic Acid (13d): Recrystallized from EtOH. Yield $84 \%$. M.p. 168 and $172^{\circ} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right): 8.88(d, J=7.2, \mathrm{NH}) ; 6.94(m, 3$ arom. H$) ; 5.22(d, J=7.2, \mathrm{CH})$; $4.15\left(s, \mathrm{CH}_{2} \mathrm{Cl}\right) ; 3.73(s, 2 \mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{ClNO}_{5}$ (287.70): C 50.10, H 4.91, N 4.87; found: C 50.4, H 4.9, N 4.65.

3,4-Dimethoxy- $\alpha$ - $/(4$-nitrobenzoyl)aminolbenzeneacetic Acid (13e): Recrystallized from EtOH. Yield $64 \%$. M.p. $187^{\circ} . \Delta H=39.5 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right): 9.32(d, J=7, \mathrm{NH}) ; 8.30,8.11\left(2 d, \mathrm{NO}_{2} \mathrm{C}_{6} \mathrm{H}_{4}\right) ; 7.11(d, 1$ arom. $\mathrm{H}) ; 6.99(m, 2$ arom. H$) ; 5.49(d, J=7, \mathrm{CH}) ; 3.75,3.74(2 s, 2 \mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{7}(360.33)$ : C 56.67, H 4.48, N 7.77; found: C 56.1, H 4.7, N 7.8.

3,4-Dimethoxy- $\alpha$-/(2-nitrobenzoyl)amino/benzeneacetic Acid (13f): Recrystallized from $\mathrm{AcOH} / \mathrm{H}_{2} \mathrm{O}$. Yield $66 \%$. M.p. 204.5 $. \Delta H=49.8 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right): 9.40(d, J=8, \mathrm{NH}) ; 8.04,8.11\left(2 \mathrm{~m}, \mathrm{NO}_{2} \mathrm{C}_{6} \mathrm{H}_{4}\right) ; 7.04$ $\left(s, 1\right.$ arom. H); $6.96\left(m, 2\right.$ arom. H); $5.45(d, J=8, \mathrm{CH}) ; 3.75,3.74(2 \mathrm{~s}, 2 \mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{7}$ (360.33): C 56.67 , H 4.48, N 7.77 ; found: C 56.6, H 4.5, N 7.6.
$\alpha$-Amino-3,4-dimethoxybenzeneacetic Acid Hydrochloride $(\mathbf{1 4} \cdot \mathrm{HCl})$. A mixture of $\mathbf{1 3 b}(3.15 \mathrm{~g}, 10 \mathrm{mmol})$, conc. HCl soln. ( 50 ml ), and $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{ml})$ was refluxed for 18 h . After cooling to r.t., the precipitated benzoic acid was filtered off. The filtrate was discolored with activated carbon (Norit) and evaporated. The residue was recrystal-
lized from $\mathrm{MeOH} / \mathrm{AcOEt}: \mathbf{1 4} \cdot \mathrm{HCl}(845 \mathrm{mg}, \mathbf{3 4} \%)$.
Methyl $\alpha$-Amino-3,4-dimethoxybenzeneacetate (15). A mixture of $14 \cdot \mathrm{HCl}(1.05 \mathrm{~g}, 4.24 \mathrm{mmol})$, methanesulfonic acid ( 0.55 ml ), and $\mathrm{MeOH}(20 \mathrm{ml})$ was refluxed for 18 h . After evaporation, the residue was dissolved in $\mathrm{H}_{2} \mathrm{O}$, conc. NaOH soln. was added, the mixture extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and the org. phase washed with $\mathrm{H}_{2} \mathrm{O}$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and evaporated: $760 \mathrm{mg}(80 \%)$ of 15 . This product was used in the next step without further purification.

Methyl $\alpha$ - $\{$ (Bis/(methoxycarbonyl)amino Jacetyl\}amino $\}-3,4$-dimethoxybenzeneacetic Acid (16a). A mixture of $\mathbf{1 5}$ ( $735 \mathrm{mg}, 3.27 \mathrm{mmol}$ ), bis[(methoxycarbonyl)amino]acetyl chloride ( $\mathbf{1 2 a} ; 808 \mathrm{mg}, 3.6 \mathrm{mmol}$; prepared according to [8]), $\mathrm{Et}_{3} \mathrm{~N}(485 \mathrm{mg}, 4.8 \mathrm{mmol})$, and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(6 \mathrm{ml})$ was stirred overnight at $r . t$. The org. phase was washed successively with dil. HCl soln., aq. $\mathrm{NaHCO}_{3}$ soln., and $\mathrm{H}_{2} \mathrm{O}$, dried ( $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ), and evaporated; pure $\mathbf{1 6 a}(900 \mathrm{mg}$, $67 \%$ ). An anal. sample was obtained by recrystallization from $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ /hexane. M.p. $202^{\circ}\left(10^{\circ} / \mathrm{min}\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ $\left(\mathrm{CDCl}_{3}\right): 7.54$ (br. $\left.d, \mathrm{NH}\right) ; 6.92-6.80(3$ arom. H); 5.94 (br. $d, 2 \mathrm{NH}) ; 5.53(t, J=7, \mathrm{H}-\mathrm{C}(\alpha)) ; 5.44(d, J=7$, $\mathrm{NHC} H \mathrm{NH}$ ); 3.88, 3.86, 3.73, 3.72, 3.67 ( $5 \mathrm{~s}, 5 \mathrm{MeO}$ ). Anal. calc. for $\mathrm{C}_{17} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{9}$ (413.39): C 49.39, H 5.61, N 10.16; found: C 49.4, H 5.6, N 10.3.
$\alpha, \alpha^{\prime}$-Bis (benzoylamino)-4,5-dimethoxybenzene-1,2-diacetic Acid (18b). Method A (from 13b): To a cooled ( $10-15^{\circ}$ ) suspension of $11 \mathrm{~b}(16.8 \mathrm{~g}, 84 \mathrm{mmol})$ in methanesulfonic acid ( 80 ml ) was added $\mathbf{1 3 b}(25.82 \mathrm{~g}, 80 \mathrm{mmol})$. Method B (from veratrole (10): To a cooled ( $10-15^{\circ}$ ) mixture of $10(11 \mathrm{~g}, 80 \mathrm{mmol})$ in methanesulfonic acid ( 160 ml ) was added $11 \mathrm{~b}(31.36 \mathrm{~g}, 160.8 \mathrm{mmol})$. Methods $A$ and $B$ : the reaction was carried alike as following and gave about the same yields. The mixture was stirred at $10-15^{\circ}$ for 15 min , at r.t. until the solids were dissolved (a few hours), and then left for 48 h . The mixture was poured into a mixture of ice/ $\mathrm{H}_{2} \mathrm{O}$ ( ca .600 ml ) and AcOEt $(1200 \mathrm{ml})$ with stirring. The org. phase was decanted, washed twice with $\mathrm{H}_{2} \mathrm{O}$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and then evaporated. The crude product ( $c a .40 \mathrm{~g}$ ) was recrystallized from AcOEt ( 160 ml ) to give $\mathbf{1 8 b}$ as a diastereoisomer mixture $\mathbf{1 8 b A} / \mathbf{1 8 b B}$ ca $2: 1(32 \mathrm{~g}, 81 \%)$ which can be used in the next step without further purification.

Separation of $\mathbf{1 8 b A} / \mathbf{1 8 b B}$ : Crude product obtained as above (ca. $40 \mathrm{~g}, 80 \mathrm{mmol}$ ) was treated with cyclohexanamine ( $15.8 \mathrm{~g}, 160 \mathrm{mmol}$ ) in $\mathrm{EtOH}(250 \mathrm{ml})$ to give a crystalline precipitate ( 30 g ) which was collected by filtration. This salt was suspended in EtOH ( 500 ml ) and the suspension refluxed for 15 min with stirring. After several hours at r.t., the precipitate was collected by filtration, washed with EtOH , and dried to give a 1 st crop ( 26.4 g ). The combined mother liquors were concentrated to give a 2 nd crop ( 10.92 g ). The last mother liquors were concentrated nearly to dryness, AcOEt was added to give a 3 rd crop $(4.43 \mathrm{~g})$. The 1 st and 2 nd crop was stirred in $1 \mathrm{~N} \mathrm{HCl}(5 \mathrm{ml} / \mathrm{mmol}$ of salt) for 2 h to give $\mathbf{1 8 b A} / \mathbf{1 8 b B}$ ca. 8:2 ( 26.34 g ). Acid $\mathbf{1 8 b A}(15 \mathrm{~g}, 38 \%)$ was obtained by treating this crop with default cyclohexanamine followed by decomposition of the resulting pure salt of $\mathbf{1 8 b A}$ Acid $18 \mathrm{bB}(2.18 \mathrm{~g}, 5 \%)$ was obtained by decomposition of the 3 rd crop.
meso-Acid 18bA: M.p. $233^{\circ}$ (dec.). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right)$ : Table 4; additionally: 12.98 (br. $\left.s, 2 \mathrm{CO}_{2} \mathrm{H}\right) ; 7.78$, 7.43 ( 2 m 2 PhCO ). Anal. calc. for $\mathrm{C}_{26} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{8}$ (492.49) : C 63.41, H 4.91, N 5.69 ; found: C 63.3, H 5.0, N 5.6.
rac-Acid 18bB: M.p. $242^{\circ}$ (dec.). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (( $\left.\mathrm{D}_{6}\right)$ DMSO): Table 4; additionally: 12.8 (br. s, $2 \mathrm{CO}_{2} \mathrm{H}$ ); 7.92, $7.50(2 \mathrm{~m}, 2 \mathrm{PhCO})$. Anal. calc. for $\mathrm{C}_{26} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{8}$ (492.49): C 63.41, H 4.91, N 5.69 ; found: C 63.3, H 5.0, N 5.75.

Table 4. Selected ${ }^{1} H-N M R$ Chemical Shifts $[\mathrm{ppm}]$ and Coupling Constants $[\mathrm{Hz}]$ for the Diastereoisomers A and B of Diacids 18b-f in $\left(D_{6}\right) D M S O$ and of Diesters $\mathbf{1 9 b}-\mathbf{d}$ in $\left.\mathrm{CDCl}_{3}{ }^{\mathbf{a}}\right)^{\mathbf{b}}$ )

|  | $\begin{aligned} & \mathbf{1 8 b A} \\ & \text { meso } \end{aligned}$ | $\begin{aligned} & \mathbf{1 8 b B} \\ & \text { rac } \end{aligned}$ | 18cA <br> meso | $\begin{aligned} & \mathbf{1 8 c B} \\ & \text { rac } \end{aligned}$ | 18dA | 18dB | 18eA | 18eB |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{H}-\mathrm{C}(\alpha)^{\text {c }}$ ) | 5.83 | 5.89 | 5.77 | 5.84 | 5.56 | 5.61 | 5.83 | 5.92 |
| $\mathrm{H}-\mathrm{C}(3)^{\text {d }}$ ) | 7.09 | 7.08 | 7.08 | 7.06 | 6.89 | 6.87 | 7.02 | 7.06 |
| $\mathrm{MeO}-\mathrm{C}(4)^{\text {d }}$ ) | 3.76 | 3.75 | $3.77^{\text {e }}$ ) | $3.74{ }^{\text {f }}$ ) | 3.73 | 3.73 | 3.76 | 3.76 |
| $\mathrm{NH}^{\text {c }}$ ) | 8.78 | 9.23 | 8.57 | 9.11 | 9.04 | 8.84 | 9.10 | 9.51 |
| $J(\alpha, \mathrm{NH})$ | 6.8 | 7.2 | 6.5 | 7.5 | 7.3 | 7.5 | 6.9 | 7.4 |
|  | 18fA | 18fB | $\begin{aligned} & \text { 19bA } \\ & \text { meso } \end{aligned}$ | $\begin{aligned} & \text { 19bB } \\ & \text { rac } \end{aligned}$ | 19cA <br> meso | $\begin{aligned} & \text { 19cB } \\ & \text { rac } \end{aligned}$ | 19dA | 19dB |
| $\mathrm{H}-\mathrm{C}(\alpha)^{\text {c }}$ ) | 5.95 | 5.82 | 5.90 | 5.92 | 5.80 | 5.89 | 5.68 | 5.85 |
| $\mathrm{H}-\mathrm{C}(3)^{\text {d }}$ ) | 7.00 | 7.00 | 6.87 | 6.84 | 6.86 | 6.83 | 6.73 | 6.75 |
| $\left.\mathrm{MeO}-\mathrm{C}(4)^{\mathrm{d}}\right)^{\mathrm{g}}$ ) | 3.76 | 3.75 | 3.81 | 3.86 | $3.84{ }^{\text {e }}$ ) | 3.85 ${ }^{\text {f }}$ ) | 3.87 | 3.86 |
| MeOOC $\left.{ }^{\text {d }}\right)^{\mathbf{g}}$ ) | - | - | 3.78 | 3.77 | 3.81 | 3.76 | 3.74 | 3.80 |
| $\mathrm{NH}^{\text {c }}$ ) | 9.40 | 9.57 | 7.72 | $8.9{ }^{\text {h }}$ ) | 7.63 | $8.8{ }^{\text {h }}$ ) | 8.68 | 7.57 |
| $J(\alpha, N H)$ | 7.3 | 7.5 | 4.8 | 5.5 | 4.5 | 5.9 | 6.0 | 6.3 |

a) A and $\mathbf{B}$ designate the two diastereoisomeric forms of 18 and 19.
${ }^{\text {b }}$ ) Resonances for other protons, see Exper. Part.
${ }^{c}$ ) $d$
${ }^{\text {d }} s$.
${ }^{\text {e }}$ ) Superimposed by the $4-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ signal.
${ }^{\text {f }}$ ) Additional $s$ for $4-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ at 3.80 ppm . Assignments may be interchanged.
${ }^{8}$ ) Assignments may be interchanged.
${ }^{\text {h }}$ ) The $d$ appears as br. $s$.

4,5-Dimethoxy- $\alpha, \alpha^{\prime}$-bis/(4-methoxybenzoyl)aminoJbenzene-1,2-diacetic Acid (18c). Method A (from 13c): As described for $\mathbf{1 8 b}$ from $11 \mathrm{c}(23.63 \mathrm{~g}, 0.105 \mathrm{~mol}), 13 \mathrm{c}(34.5 \mathrm{~g}, 0.1 \mathrm{~mol})$, and methanesulfonic acid ( 200 ml ). The mixture was stirred at r.t. until the solids were dissolved and then left for 2 days. Method $B$ (from 10). To a cooled ( $10-15^{\circ}$ ) mixture of $\mathbf{1 0}(13.8 \mathrm{~g}, 0.1 \mathrm{~mol})$ and methanesulfonic acid ( 160 ml ) was added $11 \mathrm{c}(23 \mathrm{~g}, 0.102 \mathrm{~mol})$ with stirring. The mixture was stirred at r.t. until the solids were dissolved and then left. After 2 days, $11 \mathrm{c}(23 \mathrm{~g}$, 0.102 mol ) and methanesulfonic acid ( 90 ml ) were added. The mixture was stirred at r.t. until the solids were dissolved and then left again for 2 days. Methods $A$ and $B$ : the reaction was carried alike as following. The mixture was poured into a mixture of ice $/ \mathrm{H}_{2} \mathrm{O}(c a .1: 1)$ and $\mathrm{AcOEt}(750 \mathrm{ml})$ with stirring. The org. phase was decanted, the aq. phase extracted again with $\mathrm{AcOEt}(250 \mathrm{ml})$, the combined org. phase washed twice with $\mathrm{H}_{2} \mathrm{O}$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and evaporated, and the residue recrystallized from $\mathrm{AcOEt}(100 \mathrm{ml}): 28-31 \mathrm{~g}$ of $\mathbf{1 8 c}$ (diastereoisomers A and $\mathbf{B}$ ca. 2:1) containing a small amount of $\mathbf{1 3 c}$. This product can be used in the next step without further purification.

Separation of $\mathbf{1 8 c A} / \mathbf{1 8 c B}$ : Crude product ( $30.3 \mathrm{~g}, 55 \mathrm{mmol}$ ) obtained as above was treated by cyclohexanamine ( $11 \mathrm{~g}, 110 \mathrm{mmol}$ ) in $\mathrm{EtOH}(110 \mathrm{ml})$. The resulting salt was collected by filtration and suspended in EtOH ( 10 ml per g of salt) and the suspension refluxed for 15 min with stirring. After several hours at r.t., the precipitate was collected by filtration, washed with EtOH, and dried. The operation was repeated again to give pure salt of $18 \mathrm{cA}(24 \mathrm{~g})$. The combined mother liquors were concentrated to give a 2 nd crop ( 2.33 g ) which was recrystallized from EtOH ( 23 ml ) to give pure salt of $\mathbf{1 8 c B}(1.9 \mathrm{~g})$. Treatment of these salts with $1 \mathrm{~N} \mathrm{HCl}(5 \mathrm{ml} / \mathrm{mmol}$ of salt) for 2 h with stirring gave acid 18 cA or 18 cB , resp.
meso-Acid 18cA: M.p. $217^{\circ}$ (dec.; $10^{\circ} / \mathrm{min}$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right)$ : Table 4; additionally: 13.0 (br. s , $\left.2 \mathrm{CO}_{2} \mathrm{H}\right) ; 7.76,6.88\left(2 \mathrm{MeOC}_{6} H_{4}\right)$. Anal. calc. for $\mathrm{C}_{28} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{10}(552.55)$ : C 60.87 , H 5.11 , N 5.07 ; found: C 60.8 , H 5.15, N 5.0 .
rac-Acid 18cB: M.p. $199^{\circ}$ (dec.). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right.$ ): Table 4; additionally: 12.8 (br. $s, 2 \mathrm{CO}_{2} \mathrm{H}$ ); 7.92, $6.08\left(2 \mathrm{MeOC}_{6} \mathrm{H}_{4}\right)$. Anal. calc. for $\mathrm{C}_{28} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{10} \cdot 1.5 \mathrm{H}_{2} \mathrm{O}$ (579.58): C 58.03, H 5.39, N 4.83; found: C 58.3, H 5.4, N 4.9 .
$\alpha, \alpha^{\prime}$-Bis/(chloroacetyl)amino ]-4,5-dimethoxybenzene-1,2-diacetic Acid (18d). As described for 18b, from $11 \mathbf{d}$ $(1.75 \mathrm{~g}, 10.5 \mathrm{mmol}), 13 \mathrm{~d}(2.87 \mathrm{~g}, 10 \mathrm{mmol})$, and methanesulfonic acid ( 10 ml ). After the reaction ( 48 h ), the mixture was poured into a stirred mixture of ice/ $\mathrm{H}_{2} \mathrm{O}(c a .100 \mathrm{ml}), \mathrm{NaCl}(15 \mathrm{~g})$, and $\mathrm{AcOEt}(50 \mathrm{ml})$. The org. phase was decanted, the aq. phase extracted again with $\operatorname{AcOEt}(50 \mathrm{ml})$, and the combined org. phase washed with brine $(50 \mathrm{ml})$ and then concentrated to $c a .50 \mathrm{ml}$ while crystallization occurred. After several hours at r.t., the crystalline precipitate was collected by filtration and dried; $\mathbf{1 8 d A} / \mathbf{1 8 d B}(3.45 \mathrm{~g})$ contaminated with a small amount of unidentified impurities. This product was used in the next step without further purification.

Separation of $\mathbf{1 8 d A} / \mathbf{1 8 d B}$ : To the crude product obtained as above ( $1 \mathrm{~g}, 2.29 \mathrm{mmol}$ ) $\mathrm{MeOH}(10 \mathrm{ml})$ was added, and the suspension refluxed for 10 min and then left a $-20^{\circ}$ overnight to give crystals of pure diastereoisomer $18 \mathrm{dA}(194 \mathrm{mg})$. The mother liquors were evaporated and treated with cyclohexanamine ( $0.25 \mathrm{~g}, 2.5 \mathrm{mmol}$ ) in $\mathrm{EtOH}(3 \mathrm{ml})$. The resulting salt was collected by filtration and recrystallized from $\mathrm{EtOH}(5 \mathrm{ml})$ to give pure salt of $\mathbf{1 8 d B}(110 \mathrm{mg})$. Alternately, the same crude product $(1 \mathrm{~g}, 2.29 \mathrm{mmol})$ was treated directly with cyclohexanamine $(0.39 \mathrm{~g}, 3.9 \mathrm{mmol})$ in EtOH ( 5 ml ). The resulting salt was collected by filtration $(820 \mathrm{mg})$ and recrystallized from $\mathrm{EtOH}(20 \mathrm{ml})$ to give pure salt of $18 \mathrm{dA}(410 \mathrm{mg})$. The salts were decomposed with $1 \mathrm{~N} \mathrm{HCl}(5 \mathrm{ml} / \mathrm{mmol}$ of salt) and acids 18dA or $\mathbf{1 8 d B}$ extracted with AcOEt.

Isomer 18dA: M.p. $229^{\circ}\left(\mathrm{dec} . ; 10^{\circ} / \mathrm{min}\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right):$ Table 4; additionally: 12.96 (br. $\left.s, 2 \mathrm{CO}_{2} \mathrm{H}\right)$; 4.15, 4.11 ( $\left.A B, J=13.1,2 \mathrm{CH}_{2} \mathrm{Cl}\right)$. Anal. calc. for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{8} \cdot 1.5 \mathrm{H}_{2} \mathrm{O}(464.26)$ : C 41.22, H 4.56, N 6.01 ; found: C 41.4, H 4.6, N 6.0. Cyclohexanamine Salt: Anal. calc. for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{8} \cdot 2 \mathrm{C}_{6} \mathrm{H}_{13} \mathrm{~N}(635.59)$ : C 52.91, H 6.98, N 8.82; found: C $53.0, \mathrm{H} 7.0, \mathrm{~N} 8.8$.

Isomer 18dB:M.p. $200^{\circ}$ (dec.; $10^{\circ} / \mathrm{min}$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right)$ : Table 4; additionally: 13.1 (br. $\left.s, 2 \mathrm{CO}_{2} \mathrm{H}\right)$; 4.11, ( $A B, J=13.0,2 \mathrm{CH}_{2} \mathrm{Cl}$ ). Anal. calc. for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{\mathrm{B}}(437.24)$ : C 43.95, H 4.15, N 6.41 ; found: C 44.0, H 4.26, N 6.4. Cyclohexanamine Salt: Anal. calc. for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{8} \cdot 2 \mathrm{C}_{6} \mathrm{H}_{13} \mathrm{~N}$ (635.59): C 52.91, H 6.98, N 8.82; found: C 52.8 , H 7.1, N 8.7.

4,5-Dimethoxy- $\alpha, \alpha^{\prime}$-bis[(4-nitrobenzoyl)amino]benzene-1,2-diacetic Acid (18e). As described for 18c, from crude $11 \mathrm{e}(20.5 \mathrm{~g})$, $13 \mathrm{e}(18 \mathrm{~g}, 50 \mathrm{mmol})$, and methanesulfonic acid ( 150 ml ). The crude $18 \mathrm{e}(33.5 \mathrm{~g})$ was purified via its cyclohexanamine salt as described for 18 c to give 18 eA (salt: 12 g ; acid: 7.1 g after recrystallization from AcOEt, $24 \%$ ) and $\mathbf{1 8 e B}$ (salt: 5.68 g ; acid: 3.9 g after recrystallization from $\mathrm{AcOEt}, 13 \%$ ).

Isomer 18eA: M.p. $202^{\circ}$ (dec.). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right)$ : Table 4; additionally: 8.13, $7.93\left(2 \mathrm{NO}_{2} \mathrm{C}_{6} \mathrm{H}_{4}\right)$. Anal. calc. for $\mathrm{C}_{26} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{12} \cdot 1 \mathrm{H}_{2} \mathrm{O}(600.50)$ : C $52.00, \mathrm{H} 4.03, \mathrm{~N} 9.33$; found: C 52.4, H 3.8, N 9.1 . Cyclohexanamine Salt: Anal. calc. for $\mathrm{C}_{26} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{12} \cdot 2 \mathrm{C}_{6} \mathrm{H}_{13} \mathrm{~N} \cdot 2 \mathrm{H}_{2} \mathrm{O}$ (816.86): C 55.87, H 6.42, N 10.29 ; found: C 55.8, H 6.4, N 10.3 .

Isomer 18eB: M.p. $220^{\circ}$ (dec.). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right)$ : Table 4; additionally: 8.30, $8.12\left(2 \mathrm{NO}_{2} \mathrm{C}_{6} \mathrm{H}_{4}\right)$. Anal. calc. for $\mathrm{C}_{26} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{12} \cdot 1 \mathrm{H}_{2} \mathrm{O}(600.50)$ : C $52.00, \mathrm{H} 4.03, \mathrm{~N} 9.33$; found: C $52.1, \mathrm{H} 4.0, \mathrm{~N} 9.1$. Cyclohexanamine Salt: Anal. calc. for $\mathrm{C}_{26} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{12} \cdot 2 \mathrm{C}_{6} \mathrm{H}_{13} \mathrm{~N}(780.23$ ): C $58.45, \mathrm{H} 6.20, \mathrm{~N} 10.76$; found: C $58.5, \mathrm{H} 6.3, \mathrm{~N} 10.7$.

4,5-Dimethoxy- $\alpha, \alpha^{\prime}$-bis [(2-nitrobenzoyl)amino]benzene-1,2-diacetic Acid (18f). As described for 18 c , from crude 1 If ( 5 g ), $13 \mathrm{f}(6.48 \mathrm{~g}, 18 \mathrm{mmol})$, and methanesulfonic acid ( 40 ml ). The crude $\mathbf{1 8 f}(9.17 \mathrm{~g})$ was purified via
 isomer $\mathbf{1 8 f B}$ (acid: 1.3 g after recrystallization from $\mathrm{AcOH}, 12 \%$ ).

Isomer 18fA: M.p. $223^{\circ}$ (dec.; $10^{\circ} / \mathrm{min}$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right)$ : Table 4 additionally: 8.02, $7.70\left(2 \mathrm{NO}_{2}-\right.$ $\mathrm{C}_{6} \mathrm{H}_{4}$ ). Anal. calc. for $\mathrm{C}_{26} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{12} \cdot 1 \mathrm{H}_{2} \mathrm{O}(600.50)$ : C 52.00 , H 4.03, N 9.33 ; found: C 52.3, H 4.0, N 9.4.

Isomer 18fB: M.p. $231^{\circ}$ (dec.; $10^{\circ} / \mathrm{min}$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right)$ : Table 4 additionally: $8.00,7.70$ ( $2 \mathrm{NO}_{2} \mathrm{C}_{6} \mathrm{H}_{4}$ ). Anal. calc. for $\mathrm{C}_{26} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{12}$ (582.48): C 53.61, H 3.81, N 9.62 ; found: C 53.6, H 3.9, N 9.6.

Dimethyl $\alpha, \alpha^{\prime}$-Bis (benzoylamino)-4,5-dimethoxybenzene-1,2-diacetate (19b). Method A: A mixture of $\mathbf{1 8 b}(9 \mathrm{~g}$, $18.3 \mathrm{mmol}), \mathrm{H}_{2} \mathrm{SO}_{4}(2.1 \mathrm{ml})$, and $\mathrm{MeOH}(600 \mathrm{ml})$ was refluxed for 17 h . The resulting soln. was concentrated and then diluted with $\mathrm{H}_{2} \mathrm{O}$. The crystalline precipitate was collected by filtration, washed successively with aq. $\mathrm{NaHCO}_{3}$ soln. and $\mathrm{H}_{2} \mathrm{O}$, and then air-dried at $70^{\circ}: 19 \mathrm{~b}(8.1 \mathrm{~g}, 85 \%)$. Method B: A mixture of $\mathbf{1 8 b}(19 \mathrm{~g}$, $38.6 \mathrm{mmol}), \mathrm{H}_{2} \mathrm{SO}_{4}(0.3 \mathrm{ml}), \mathrm{MeOH}(12 \mathrm{ml})$, and 1,2 -dichloroethane ( 80 ml ) was refluxed for $17-24 \mathrm{~h}$. After cooling to r.t., the org. phase was washed with $\mathrm{H}_{2} \mathrm{O}$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and then filtered through a short column of $\mathrm{Al}_{2} \mathrm{O}_{3}\left(80 \mathrm{~g}, \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 1: 99\right) ; 19 \mathrm{~b}(17.1 \mathrm{~g}, 85 \%)$.
meso-Ester 19bA: Recrystallized from AcOEt. M.p. $185.5^{\circ} . \Delta H=52.0 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 4; additionally: $7.88,7.44(2 m, 2 \mathrm{PhCO}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 171.0\left(\mathrm{CO}_{2} \mathrm{Me}\right) ; 166.6(\mathrm{CONH}) ; 149.4$ ( C - OMe ); 133.12, 131.6 (quat. C); $128.25,127.1,110.9(\mathrm{CH}) ; 55.85(\mathrm{MeO}) ; 54.3,52.8(\mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{28} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{8}(520.55)$ : C 64.61, H 5.42, N 5.38 ; found: C 64.4, H 5.2, N 5.4 .
rac-Ester 19bB: Recrystallized from AcOEt. M.p. $219^{\circ} . \Delta H=51.0 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 4 ; additionally: $8.01,7.50(2 m, 2 \mathrm{PhCO})$. Anal. calc. for $\mathrm{C}_{28} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{8}(520.55)$ : C 64.61, H 5.42, N 5.38 ; found: C 64.6, H 5.3, N 5.3 .

Dimethyl 4,5-Dimethoxy- $\alpha, \alpha^{\prime}$-bis/(4-methoxybenzoyl)amino/benzene-1,2-diacetate (19c). As described for 19b.
meso-Ester 19cA : Recrystallized from $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$. M.p. $222^{\circ} . \Delta H=66.6$. Chromatography on a chiral phase (DNB PHGly 2.4.3, $4.6 \times 250 \mathrm{~mm}$, heptane $/ \mathrm{EtOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 75: 1: 24$, flow rate $1 \mathrm{ml} / \mathrm{min}$, UV detection at $254 \mathrm{~nm}): 1$ peak, $t_{\mathrm{R}} 34 \mathrm{~min} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 4; additionally: 7.87, $6.91\left(2 \mathrm{MeOC}_{6} \mathrm{H}_{4}\right)$. Anal. calc. for $\mathrm{C}_{30} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{O}_{10}$ ( 580.60 ): C $62.06, \mathrm{H} 5.56, \mathrm{~N} 4.83$; found: C $62.3, \mathrm{H} 5.6, \mathrm{~N} 4.8$.
rac-Ester 19cB: Recrystallized from MeOH. M.p. 191.5 $5^{\circ} . \Delta H=36.3$. Chromatography on a chiral phase (see 19cA): 2 peaks, $t_{\mathrm{R}} 26$ and 30 min . ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 4; additionally: 7.98, $6.92\left(2 \mathrm{MeOC}_{6} H_{4}\right)$. Anal. calc. for $\mathrm{C}_{30} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{O}_{10}$ ( 580.60 ): C $62.06, \mathrm{H} 5.56, \mathrm{~N} 4.83$; found: $\mathrm{C} 62.1, \mathrm{H} 5.5, \mathrm{~N} 4.65$.

Dimethyl a, $\alpha^{\prime}$-Bis/(chloroacetyl) amino ]-4,5-dimethoxybenzene-1,2-diacetate (19d). To a soln. of crude 18dA/ $\mathbf{1 8 d B}(2.28 \mathrm{~g} ; c a .2: 1)$ in $\mathrm{MeOH}(60 \mathrm{ml})$ was added portionwise a soln. of diazomethane in $\mathrm{Et}_{2} \mathrm{O}(80 \mathrm{ml})$. After 1 h , the $\mathrm{Et}_{2} \mathrm{O}$ was evaporated at r.t. The residue was purified by $\mathrm{CC}\left(\mathrm{SiO}_{2}, 100 \mathrm{~g}\right.$, acetone $/ \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ $2.5: 2.5: 95$ ) to give successively pure $19 \mathrm{dA}(400 \mathrm{mg}), 19 \mathrm{dA} / 19 \mathrm{~dB}(600 \mathrm{mg})$, and pure $19 \mathrm{~dB}(250 \mathrm{mg})$.

Isomer 19dA: M.p. $172^{\circ} . \Delta H=49.3 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 4; additionally: 4.09, 4.06 ( $A B, J=14.6$, $2 \mathrm{CH}_{2} \mathrm{Cl}$ ). Anal. calc. for $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{8}$ (465.29): C 46.46, H 4.77, N 6.02; found: C 46.8, H 4.75, N 6.0 .

Isomer 19dB: M.p. 166 and $172^{\circ} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 4; additionally: 4.08, $4.06(\mathrm{AB}, J=15.0$, $2 \mathrm{CH}_{2} \mathrm{Cl}$ ). Anal. calc. for $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O}_{8}$ (465.29): C $46.46, \mathrm{H} 4.77, \mathrm{~N} 6.02$; found: C $46.5, \mathrm{H} \mathrm{4.8}, \mathrm{~N} 5.9$.

Methyl 4-Amino-1,2,3,4-tetrahydro-6,7-dimethoxy-3-oxoisoquinoline-1-carboxylate (6a). The procedure described in [10] was modified as follows: A mixture of $19 \mathrm{~d}(\mathbf{A} / \mathbf{B}$ ca. $1: 1 ; 3.4 \mathrm{~g}, 7.31 \mathrm{mmol}$ ), $N$-piperidinothiourea [28] ( $3.9 \mathrm{~g}, 27.1 \mathrm{mmol}$ ), $\mathrm{AcOH}(0.35 \mathrm{ml})$, and abs. $\mathrm{EtOH}(150 \mathrm{ml})$ was refluxed for 4 h under $\mathrm{N}_{2}$. After evaporation, the residue was dissolved in $\mathrm{CHCl}_{3}(300 \mathrm{ml})$ and the org. phase washed with $\mathrm{H}_{2} \mathrm{O}(2 \times 50 \mathrm{ml})$. To the combined aq. phase was added conc. $\mathrm{NH}_{4} \mathrm{OH}$ soln. The resulting cloudy mixture was extracted twice with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The org. phase was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated at r.t. and the residue recrystallized from AcOEt : $6 \mathrm{a}\left(0.9 \mathrm{~g}, 44 \%\right.$ ) as a 8:2 diastereoisomer mixture. Anal. calc. for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{5}$ (280.29): C 55.71, H 5.75, N 9.99; found: C 55.5, H 5.7, N 10.0 .

Major Isomer: ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 7.31(s, \mathrm{H}-\mathrm{C}(5)$ or $\mathrm{H}-\mathrm{C}(8)) ; 6.86(s, \mathrm{H}-\mathrm{C}(8)$ or $\mathrm{H}-\mathrm{C}(5)) ; 6.72(d, J=5$, $\mathrm{NH}) ; 5.03(d, J=5, \mathrm{H}-\mathrm{C}(1)) ; 4.54(s, \mathrm{H}-\mathrm{C}(4)) ; 3.92,3.90,3.71$ ( $3 s, 3 \mathrm{MeO}$ ).

Minor Isomer: ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 7.01(s, \mathrm{H}-\mathrm{C}(5)$ or $\mathrm{H}-\mathrm{C}(8)) ; 6.83(s, \mathrm{H}-\mathrm{C}(8)$ or $\mathrm{H}-\mathrm{C}(5)$ ); 6.6 (br. $d$, $\mathrm{NH}) ; 5.11(d, J=3, \mathrm{H}-\mathrm{C}(1)) ; 4.24(s, \mathrm{H}-\mathrm{C}(4)) ; 3.91,3.89,3.80(3 s, 3 \mathrm{MeO})$.

Methyl 3-Hydroxy-6,7-dimethoxyisoquinoline-1-carboxylate (20). A mixture of $\mathbf{6 a}(145 \mathrm{mg}, 0.52 \mathrm{mmol})$ and $\mathrm{MeOH}(100 \mathrm{ml})$ was refluxed under $\mathrm{N}_{2}$ for 1.5 h to give a bright yellow soln. The solvent was evaporated, the residue taken in acetone, and the soln. filtrated through a short column of $\mathrm{SiO}_{2}(7 \mathrm{~g})$ : pure $\mathbf{2 0}(125 \mathrm{mg}, 91 \%)$ as yellow crystals. An anal. sample was obtained by recrystallization from $\mathrm{CHCl}_{2} / \mathrm{Et}_{2} \mathrm{O}$. M.p. $222^{\circ}\left(10^{\circ} / \mathrm{min}\right)$ ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 8.06(s, \mathrm{H}-\mathrm{C}(8)) ; 7.09(s, \mathrm{H}-\mathrm{C}(4)) ; 6.78(s, \mathrm{H}-\mathrm{C}(5)) ; 3.99,3.98,3.95(3 s, 3 \mathrm{MeO}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ $\left(\mathrm{CDCl}_{3}\right): 164.3\left(\mathrm{CO}_{2} \mathrm{Me}\right) ; 158.8(\mathrm{C}(3)) ; 154.1(\mathrm{C}(6)) ; 150.3(\mathrm{C}(7)) ; 140.5(\mathrm{C}(4 \mathrm{a})) ; 136.4(\mathrm{C}(1)) ; 119.1(\mathrm{C}(8 \mathrm{a})) ; 110.2$ (C(4)); 103.1, $102.7(\mathrm{C}(8), \mathrm{C}(5)) ; 55.9,55.8(\mathrm{MeO}) ; 52.6\left(\mathrm{CO}_{2} \mathrm{Me}\right)$. Anal. calc. for $\mathrm{C}_{13} \mathrm{H}_{13} \mathrm{NO}_{5}(280.28): \mathrm{C} 59.31$, H 4.98, N 5.32 ; found: C 59.5 , H 5.0, N 5.1.

Dimethyl 4,5-Dimethoxy- $\alpha, \alpha^{\prime}$-bis[(phenylthioxomethyl)aminolbenzene-1,2-diacetate (21b). To a soln. of 19bA ( $3.1 \mathrm{~g}, 6 \mathrm{mmol}$ ) in dry THF ( 240 ml ) was added Lawesson's reagent ( $5 \mathrm{~g}, 12.4 \mathrm{mmol}$ ). The mixture was stirred at $60^{\circ}$ for $18-24 \mathrm{~h}$ and then evaporated. The residue was purified by $\mathrm{CC}\left(\mathrm{SiO}_{2}, 300 \mathrm{~g}, \mathrm{AcOEt} /\right.$ hexane $\left.1: 1\right)$ to give $\mathbf{2 1 b A}(1.65 \mathrm{~g})$ followed by a mixture of 21bA and dimethyl $\alpha$-(benzoylamino)-4,5-dimethoxy- $\alpha^{\prime}$-[(phenylthioxo-methyl)aminolbenzene-1,2-diacetate. A second CC ( $\left.\mathrm{SiO}_{2}, 100 \mathrm{~g}\right)$ gave an additional crop of $\mathbf{2 1 b A}(0.6 \mathrm{~g})$. Overall yield $68 \%$. M.p. $208^{\circ} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 8.53(d, J=5.5,2 \mathrm{NH}) ; 7.84,7.44(2 \mathrm{~m}, 2 \mathrm{PhCS}) ; 6.90(\mathrm{~s}, 2$ arom. H); $6.42(d, J=5.5,2 \mathrm{CH}) ; 3.86,3.83(s, 4 \mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{28} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{6} \mathrm{~S}_{2}$ (552.67): C 60.85, H 5.11, N 5.07 ; found: C 60.8, H 5.2, N 5.0 .

Dimethyl 4,5-Dimethoxy- $\alpha, \alpha^{\prime}$-bis $\{/(4$-methoxyphenyl) thioxomethyllamino\}benzene-1,2-diacetate (21c). Representative procedure for 21 cA : To a soln. of $19 \mathrm{cA}(7 \mathrm{~g}, 12 \mathrm{mmol})$ in dry THF ( 400 ml ) was added Lawesson's reagent ( $14 \mathrm{~g}, 34.6 \mathrm{mmol}$ ). The mixture was stirred at $60^{\circ}$ for $18-24 \mathrm{~h}$ and the solvent evaporated. The residue was purified by $\mathrm{CC}\left(\mathrm{SiO}_{2}, 200 \mathrm{~g}, \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 0: 100\right.$ to $\left.3: 97\right): 21 \mathrm{cA}(5.7 \mathrm{~g}, 77 \%)$.
meso-Ester 21cA: Recrystallized from $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{Et}_{2} \mathrm{O}$. M.p. $214^{\circ}$ (dec.). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 8.46(d, J=5$, $2 \mathrm{NH}) ; 7.86,6.88\left(2 \mathrm{MeOC}_{6} H_{4}\right) ; 6.89(s, 2$ arom. H$) ; 6.40(d, J=5,2 \mathrm{CH}) ; 3.85,3.84,3.82(3 s, 6 \mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{30} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{O}_{8} \mathrm{~S}_{2}$ (612.57): C 58.81, H 5.26, N 4.57 ; found: $\mathrm{C} 58.9, \mathrm{H} 5.2, \mathrm{~N} 4.3$.
rac-Ester 21cB: Recrystallized from $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{Et}_{2} \mathrm{O} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 8.69$ (br. d, NH); 7.71, 6.78 $\left(2 \mathrm{MeOC}_{6} H_{4}\right) ; 6.93(s, 2$ arom. H) $; 6.50(d, J=6.5,2 \mathrm{CH}) ; 3.88,3.80,3.77(3 s, 6 \mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{30} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{O}_{8} \mathrm{~S}_{2}$ (612.57): C 58.81, H 5.26, N 4.57 ; found: C 58.7 , H 5.35, N 4.4.

Preparation of 2'a and 2'b: General Procedure. Raney-Ni ( 125 g ), washed with $\mathrm{H}_{2} \mathrm{O}(3 \times)$ and then with $\mathrm{MeOH}(3 \times)$, was added to yellow soln. of 21b, $\mathbf{c}$ (either diastereoisomer $\mathbf{A}$ or $\mathbf{B}$ or their mixtures, $8-10 \mathrm{mmol}$ ) in a mixture of THF ( 200 ml ) and $\mathrm{MeOH}(200 \mathrm{ml})$. The mixture was vigorously stirred for 8 min , the Ni filtered off and rinsed with MeOH , and the colorless filtrate evaporated at $\mathbf{r}$.t. The resulting crude $\mathbf{2 2 b} \mathbf{b}, \mathbf{c}$ was dissolved in $\mathrm{CHCl}_{3}$ ( 400 ml ) and the soln, left at r.t. for 48 h . After evaporation, the residue (crude $\mathbf{2 3 b}, \mathrm{c}$ ) was dissolved in $\mathrm{MeOH}(300 \mathrm{ml})$ and treated with $1 \mathrm{~m} \mathrm{NaOMe}(22 \mathrm{ml})$. After 24 h at r.t., aq. $1 \mathrm{M} \mathrm{AcOH}(22 \mathrm{ml})$ was added and the mixture evaporated at r.t. The residue was purified by $\mathrm{CC}\left(\mathrm{SiO}_{2}, 25 \mathrm{~g}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ followed by recrystallization from $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{Et}_{2} \mathrm{O}$. Overall yield from 21b, c: $50-60 \%$ of 2 'a, b.

Dimethyl meso- $\alpha$, $\alpha^{\prime}$-Bis (benzylamino)-4,5-dimethoxybenzene-1,2-diacetate (22bA): crude product obtained from 21bA (meso). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 7.29(\mathrm{~m}, 2 \mathrm{Ph}) ; 6.91(s, 2$ arom. H$) ; 4.80\left(s, \mathrm{H}-\mathrm{C}(\alpha), \mathrm{H}-\mathrm{C}\left(\alpha^{\prime}\right)\right) ; 3.86(s$, 2 MeO ); $3.67\left(s, 2 \mathrm{CO}_{2} \mathrm{Me}\right) ; 3.73,3.63\left(A B, J=9.5,2 \mathrm{PhCH}_{2}\right)$.

Methyl trans-2-Benzyl-4-(benzylamino)-1,2,3,4-tetrahydro-6,7-dimethoxy-3-oxoisoquinoline-1-carboxylate (23bA): crude product obtained 22bA (meso). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : $7.25-7.48(12 \mathrm{H}, \mathrm{NH}, \mathrm{H}-\mathrm{C}(5)$ or $\mathrm{H}-\mathrm{C}(8)$, Ph ); $6.72(s, \mathrm{H}-\mathrm{C}(8)$ or $\mathrm{H}-\mathrm{C}(5)) ; 5.25,4.32\left(A B, J=15.0, \mathrm{CH}_{2}-\mathrm{N}(2)\right) ; 4.90(s, \mathrm{H}-\mathrm{C}(1)) ; 4.59(s, \mathrm{H}-\mathrm{C}(4)) ; 3.86(s$, $\mathrm{CH}_{2} \mathrm{NH}-\mathrm{C}(4)$, $\mathrm{MeO}-\mathrm{C}(6)$ or $\left.\mathrm{MeO}-\mathrm{C}(7)\right) ; 3.84(s, \mathrm{MeO}-\mathrm{C}(7)$ or $\mathrm{MeO}-\mathrm{C}(6)) ; 3.56\left(s, \mathrm{CO}_{2} \mathrm{Me}\right)$.

Methyl trans-1,2,3,4-Tetrahydro-6,7-dimethoxy-2-(4-methoxybenzyl)-4-[(4-methoxybenzyl) amino 3 -3-oxoiso-quinoline-1-carboxylate (23cA): crude product obtained from 21cA (meso). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 7.37,7.17,6.90$, $6.86\left(2 \mathrm{MeOC}_{6} H_{4}\right) ; 7.35(\mathrm{~s}, \mathrm{H}-\mathrm{C}(5)$ or $\mathrm{H}-\mathrm{C}(8)$ ); 6.71 ( $s, \mathrm{H}-\mathrm{C}(8)$ or $\mathrm{H}-\mathrm{C}(5)$ ); 5.14, 4.29 ( $A B, J=14.8$, $\left.\mathrm{CH}_{2}-\mathrm{N}(2)\right) ; 4.88(s, \mathrm{H}-\mathrm{C}(1)) ; 4.55(\mathrm{~s}, \mathrm{H}-\mathrm{C}(4)) ; 3.80\left(\mathrm{CH}_{2} \mathrm{NH}-\mathrm{C}(4)\right) ; 3.88,3.84,3.78(3 \mathrm{~s}, 4 \mathrm{MeO}) ; 3.56$ ( $s, \mathrm{CO}_{2} \mathrm{Me}$ ).

Methyl cis-1,2,3,4-Tetrahydro-6,7-dimethoxy-2-(4-methoxybenzyl)-4-[(4-methoxybenzyl)amino ]-3-oxoisoqui-noline-1-carboxylate (23cB): crude product obtained from 21cB (rac). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 7.35,7.19,6.87$ $\left(2 \mathrm{MeOC}_{6} H_{4}\right) ; 6.73(s, \mathrm{H}-\mathrm{C}(5)) ; 6.68(s, \mathrm{H}-\mathrm{C}(8)) ; 5.30,4.10\left(A B, J=14.8, \mathrm{CH}_{2}-\mathrm{N}(2)\right) ; 4.88(s, \mathrm{H}-\mathrm{C}(1)) ; 4.24$ ( $s, \mathrm{H}-\mathrm{C}(4)$ ); 3.96, $3.90\left(A B, J=12.7,2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{NH}-\mathrm{C}(4)\right.$ ); 3.86, 3.82, 3.79 ( $3 \mathrm{~s}, 4 \mathrm{MeO}$ ); 3.60 ( $s, \mathrm{CO}_{2} \mathrm{Me}$ ).

2,9-Dibenzyl-1,2-dihydro-6,7-dimethoxy-4,1-(iminomethano) isoquinoline-3,10(4H)-dione (2'a): M.p. 186.5 ${ }^{\circ}$. IR: $1680(\mathrm{CO}-\mathrm{N}) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 1 ; additionally: 7.21, $7.01(2 \mathrm{~m}, 2 \mathrm{~Pb}) ; 3.71(\mathrm{~s}, 2 \mathrm{MeO}) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ $\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right)$ : Table 2, additionally: 7.27, $7.09(2 \mathrm{~m}, 2 \mathrm{Ph}) ; 3.69(s, 2 \mathrm{MeO}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right):$ Table 3; additionally: 135.1, 128.5, $128.1,127.7$ (arom. C); $55.9(\mathrm{MeO}) ; 48.4\left(\mathrm{CH}_{2} \mathrm{~N}\right) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right):$ Table 3; additionally: $136.5,128.3,127.7,127.4$ (arom. CH$) ; 55.8(\mathrm{MeO}) ; 47.6\left(\mathrm{CH}_{2} \mathrm{~N}\right)$. Anal. calc. for $\mathrm{C}_{26} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{4}$ (428.47): C 72.88, H 5.65 , N 6.54 ; found: C 72.7, H $5.6, \mathrm{~N} 6.4$.

1,2-Dihydro-6,7-dimethoxy-2,9-bis(4-methoxybenzyl)-4,1-(iminomethano) isoquinoline-3,10(4H)-dione (2'b): M.p. $159^{\circ} . \Delta H=42.4$. HPLC (Chiralcel OC, EtOH, flow rate $30 \mathrm{ml} / \mathrm{h}$; UV detection at $254 \mathrm{~nm}, \alpha$ detection at $365 \mathrm{~nm}) 2$ peaks, $t_{\mathrm{R}} 21$ and $25 \mathrm{~min} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 1 . Additionally: 6.74, $6.75\left(2 \mathrm{MeOC}_{6} \mathrm{H}_{4}\right) ; 3.77,3.74$ ( $2 s, 4 \mathrm{MeO}$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right):$ Table 2; additionally: 7.02, $6.82\left(2 \mathrm{MeOC}_{6} \mathrm{H}_{4}\right) ; 3.71,3.68(2 s, 4 \mathrm{MeO})$. ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right):$ Table 3; additionally: 158.7, 129.2, 128.4, 113.8 (arom. C); 55.9, 55.1 (MeO); 47.1 $\left(\mathrm{CH}_{2} \mathrm{~N}\right)$. MS: $489.5\left([M+1]^{+}\right)$. Anal. calc. for $\mathrm{C}_{28} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{6}$ (488.55): C 68.84, H 5.78, N 5.73; found: C 68.1, H 5.6, N 5.7.

Preparation of $\mathbf{2}^{\prime} \mathbf{c}-\mathbf{e}$ : General Procedure. A mixture of $\mathbf{2}^{\prime} \mathrm{g}$ (see below; $\left.650 \mathrm{mg}, 1.5 \mathrm{mmol}\right), \mathrm{K}_{2} \mathrm{CO}_{3}(1 \mathrm{~g}$, $7.2 \mathrm{mmol})$, alkyl bromide ( 6.6 mmol ), and DMF ( 15 ml ) was stirred at $80^{\circ}$ overnight under $\mathrm{N}_{2}$. After being cooled, the mixture was diluted with $\mathrm{H}_{2} \mathrm{O}$ and the resulting precipitate collected by filtration. The purification varied slightly for $\mathbf{2}^{\prime} \mathbf{c}, \mathbf{2}^{\prime} \mathbf{d}$, and $\mathbf{2}^{\mathbf{\prime}} \mathbf{e}$ (see below).

6,7-Dibutoxy-2,9-bis(4-butoxybenzyl)-1,2-dihydro-4,1-(iminomethano) isoquinoline-3,10(4H)-dione (2'c): Purification by $\mathrm{CC}\left(\mathrm{SiO}_{2}, 30 \mathrm{~g}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 100: 0\right.$ to $98: 2$ ) afforded $\mathbf{2} \mathrm{c}$ ( $780 \mathrm{mg}, 79 \%$ ). Recrystallization from $\mathrm{CH}_{2} \mathrm{Cl}_{2} /$ pentane. M.p. $93^{\circ}, \Delta H=53.0 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 1; additionally: 6.92, $6.75\left(2 \mathrm{BuOC}_{6} \mathrm{H}_{4}\right) ; 3.87$ $\left(m, 4 \mathrm{CH}_{2} \mathrm{O}\right) ; 1.74\left(m, 4 \mathrm{CH}_{2} \mathrm{Me}\right) ; 1.45\left(m, 4 \mathrm{CH}_{2}\right) ; 0.97(t, 4 \mathrm{Me})$. Anal. calc. for $\mathrm{C}_{40} \mathrm{H}_{52} \mathrm{~N}_{2} \mathrm{O}_{6}(656.86): \mathrm{C} 73.14$, H 7.98, N 4.27; found: C 73.0, H 8.0, N 4.3.

6,7-Bis (hexyloxy)-2,9-bis [4-(hexyloxy) benzyl]-1,2-dihydro-4,1-(iminomethano) isoquinoline - 3,10 ( $4 \mathbf{H}$ )-dione ( $\mathbf{2}^{\prime} \mathrm{d}$ ): Purification by $\mathrm{CC}\left(\mathrm{SiO}_{2}, 30 \mathrm{~g}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 100: 0\right.$ to $98: 2$ ) afforded 2'd ( $\mathbf{7 8 0 \mathrm { mg } , 6 8 \% \text { ). }}$ Recrystallization from $\mathrm{CH}_{2} \mathrm{Cl}_{2} /$ pentane. M.p. $84^{\circ} . \Delta H=67.0 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 1; additionally: 6.92, $6.75\left(2(\mathrm{Hex}) \mathrm{OC}_{6} \mathrm{H}_{4}\right) ; 3.84\left(m, 4 \mathrm{CH}_{2} \mathrm{O}\right) ; 1.75\left(m, 4 \mathrm{CH}_{2} \mathrm{Me}\right) ; 1.36\left(m, 12 \mathrm{CH}_{2}\right) ; 0.91(t, 4 \mathrm{Me})$. Anal. calc. for $\mathrm{C}_{48} \mathrm{H}_{68} \mathrm{~N}_{2} \mathrm{O}_{6}$ (769.04): C 74.96, H 8.91, N 3.64; found: C 74.9, H 8.8, N 3.5.

6,7-Bis(dodecyloxy)-2,9-bis 4 -(dodecyloxy)benzyl/-1-2-dihydro-4,1-(iminomethano)isoquinoline-3,10(4H)dione ( $\mathbf{2}^{\prime} \mathbf{e}$ ). Recrystallization from dioxane gave a 1 st crop ( 755 mg ). CC of the mother liquors ( $\mathrm{SiO}_{2}, 30 \mathrm{~g}$, benzene/THF 95:5) followed by recrystallization from THF/MeOH gave a 2nd crop ( 300 mg ). Overall yield of $2^{\prime} \mathrm{e}: ~ 63 \%$. M.p. 93 and $103^{\circ} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right)\right.$ THF): Table 1; additionally: 7.09, $6.87\left(2 \mathrm{ROC}_{6} \mathrm{H}_{4}\right) ; 3.98(m$,
$\left.4 \mathrm{CH}_{2} \mathrm{O}\right)$; $1.85\left(m, 4 \mathrm{CH} \mathrm{C}_{2} \mathrm{Me}\right) ; 1.43\left(m, 36 \mathrm{CH}_{2}\right) ; 1.02(t, 4 \mathrm{Me})$. Anal. calc. for $\mathrm{C}_{72} \mathrm{H}_{116} \mathrm{~N}_{2} \mathrm{O}_{6}(1105,72): \mathrm{C} 78.21$, H 10.57, N 2.53 ; found: C $78.2, \mathrm{H} 10.6, \mathrm{~N} 2.6$.

6,7-Bis (dodecyloxy)-1,2-dihydro-2,9-bis(4-methoxybenzyl)-4,1-(iminomethano)isoquinoline-3,10( 4 H )-dione ( $\mathbf{2}^{\prime} \mathrm{f}$ ). A mixture of $2^{\prime} \mathrm{h}$ (see below, $690 \mathrm{mg}, 1.5 \mathrm{mmol}$ ), $\mathrm{K}_{2} \mathrm{CO}_{3}(1 \mathrm{~g}, 7.2 \mathrm{mmol}$ ), 1 -bromododecane ( 0.85 g , 3.4 mmol ), and DMF ( 15 ml ) was stirred at $80^{\circ}$ overnight under $\mathrm{N}_{2}$. The mixture was cooled to r.t. and diluted with $\mathrm{H}_{2} \mathrm{O}$ and the resulting precipitate collected by filtration. $\mathrm{CC}\left(\mathrm{SiO}_{2}, 25 \mathrm{~g}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ then $\left.\mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 2: 98\right)$ followed by recrystallization from $\mathrm{CH}_{2} \mathrm{Cl}_{2} /$ pentane gave pure $2^{\prime} \mathrm{f}(585 \mathrm{mg})$. Prep. TLC ( $\mathrm{SiO}_{2}, \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ $2: 98$ ) of the mother liquors afforded a 2 nd crop ( 138 mg ). Overall yield of $\mathbf{2}^{\prime} \mathrm{f}: 60 \%$. M.p.: 2 transitions; $82^{\circ}$, $\Delta H=57.8$ (attributed to a crystal $\rightarrow$ crystal transformation); $115.5^{\circ}, \Delta H=50.3$ (crystal $\rightarrow$ isotropic liquid). ${ }^{1} \mathrm{H}$-NMR $\left(\mathrm{CDCl}_{3}\right)$ : Table 1; additionally: 6.95, $6.76\left(2 \mathrm{MeOC}_{6} \mathrm{H}_{4}\right) ; 3.80\left(m, 2 \mathrm{CH}_{2} \mathrm{O}\right) ; 3.77(s, 2 \mathrm{MeO}) ; 1.75$ $\left(m, 2 \mathrm{CH}_{2} \mathrm{Me}\right) ; 1.27\left(m, 18 \mathrm{CH}_{2}\right) ; 0.88(t, 2 \mathrm{Me})$. Anal. calc. for $\mathrm{C}_{50} \mathrm{H}_{72} \mathrm{~N}_{2} \mathrm{O}_{6}$ (797.13): C 75.34, H 9.11, N 3.51; found: C 75.2, H 9.0, N 3.4.

1,2-Dihydro-6,7-dihydroxy-2,9-bis(4-hydroxybenzyl)-4,1-(iminomethano)isoquinoline-3,10(4H)-dione (2'g): To an ice-cold soln. of $\mathbf{2}^{\prime} \mathbf{b}(1.95 \mathrm{~g}, 4 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{ml})$ was added dropwise ( 30 min ) $\mathrm{BBr}_{3}(2.4 \mathrm{ml}, 25 \mathrm{mmol})$ with stirring. Stirring was continued at r.t. for 5 h , and then $\mathrm{H}_{2} \mathrm{O}(30 \mathrm{ml})$ was cautiously added. After 45 min , the crystalline precipitate was collected by filtration and dried. Trituration with MeOH afforded $\mathbf{2}^{\prime} \mathrm{g}(1.6 \mathrm{~g}, 93 \%)$. M.p. $277^{\circ}$ (dec.; $10^{\circ} / \mathrm{min}$ ). IR: 3400, 3200, 1680, 1640, 1615, 1595, $1510 .{ }^{1} \mathrm{H}$-NMR ( $\left.\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right):$ Table 2; additionally: 9.37, $9.12(2 \mathrm{br} . \mathrm{s}, 4 \mathrm{OH}) ; 6.90,6.65\left(2 d, J=8.5,2 \mathrm{HOC}_{6} H_{4}\right)$. Anal. calc. for $\mathrm{C}_{24} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{6}(432.43)$ : C 66.66, H 4.66, N 6.48; found: C 66.5 , H 4.75 , N 6.6.

1,2-Dihydro-6,7-dihydroxy-2,9-bis(4-methoxybenzyl)-4,1-(iminomethano)isoquinoline-3,10(4 H )-dione ( $\mathbf{2}^{\prime} \mathrm{h}$ ): To a mixture of $2^{\prime} \mathrm{g}(1.3 \mathrm{~g}, 3 \mathrm{mmol}), 1 \mathrm{~N} \mathrm{NaOH}(6 \mathrm{ml})$, and $\mathrm{H}_{2} \mathrm{O}(15 \mathrm{ml})$, a soln. of borax $(2.3 \mathrm{~g}, 6 \mathrm{mmol})$ in $\mathrm{H}_{2} \mathrm{O}$ ( 30 ml ) was added. The turbid mixture was stirred at r.t. for 1 h after which time it became homogeneous. Via two syringes were simultaneously added $\mathrm{Me}_{2} \mathrm{SO}_{4}(3.5 \mathrm{ml}, 37 \mathrm{mmol})$ and $5 \mathrm{~N} \mathrm{NaOH}(3.5 \mathrm{ml})$ within 30 min . After 2.5 h stirring, the mixture was acidified with $3.6 \mathrm{NH}_{2} \mathrm{SO}_{4}(7.5 \mathrm{ml})$ and then refluxed for 15 min . After cooling to r.t., the resulting precipitate was collected by filtration, rinsed with $\mathrm{H}_{2} \mathrm{O}$, and air-dried. Trituration with MeOH gave $\mathbf{2}^{\prime} \mathrm{h}(1.22 \mathrm{~g}, 81 \%)$. M.p. $236-238^{\circ} . \Delta H=51 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right):$ Table 2; additionally: 9.12 (br. $s, 2 \mathrm{OH}$ ); 7.02, $6.82\left(2 \mathrm{~m}, 2 \mathrm{HOC}_{6} H_{4}\right) ; 3.72(\mathrm{~s}, 2 \mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{26} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{6}(460.49): \mathrm{C} 67.82, \mathrm{H} 5.25, \mathrm{~N} 6.08$; found: C 67.7, H 5.5, N 5.9.

6,7-Bis(acetyloxy)-1,2-dihydro-2,9-bis(4-methoxybenzyl)-4,1-(iminomethano)isoquinoline-3,10(4H)-dione ( $\left.\mathbf{3}^{\prime} \mathbf{b}\right)$ : A mixture of $\mathbf{2} \mathbf{h}(3 \mathrm{~g}, 6.5 \mathrm{mmol})$ and $\mathrm{Ac}_{2} \mathrm{O}$ was refluxed for 2 h . Excess $\mathrm{Ac}_{2} \mathrm{O}$ was evaporated and the residue recrystallized from $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{Et}_{2} \mathrm{O}$ to give $\mathbf{3}^{\mathbf{h}}(2.9 \mathrm{~g}, 82 \%)$. M.p. $135^{\circ} . \Delta H=41.4 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table $f$; additionally: $6.96,6.80\left(2 \mathrm{MeOC}_{6} H_{4}\right) ; 3.79(s, 2 \mathrm{MeO}) ; 2.26(s, 2 \mathrm{MeCO})$. Anal. calc. for $\mathrm{C}_{30} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{8}(544.56)$ : C 66.17, H 5.18, N 5.14; found: C $66.0, \mathrm{H} 5.1, \mathrm{~N} 5.0$.

Preparation of $\mathbf{3}^{\prime} \mathbf{a}, \mathbf{c}-\mathbf{e}$ : General Procedure. To a soln. of $\mathbf{2}^{\prime} \mathrm{h}(230 \mathrm{mg}, 0.5 \mathrm{mmol})$ in pyridine ( 2 ml ) was added excess acyl chloride (for $\mathbf{3}^{\prime} \mathrm{a}, \mathrm{e}$ ) or anhydride (for $\mathbf{3}^{\prime} \mathbf{c}, \mathbf{d}$ ). The mixture was stirred at r.t. overnight, diluted with $\mathrm{H}_{\mathbf{2}} \mathrm{O}$ and then extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The org. phase was washed with dil. HCl soln. ( $3 \times$ ), and $\mathrm{H}_{2} \mathrm{O}$, dried, and evaporated. The residue was purified by TLC ( $\mathrm{SiO}_{2}, 2 \mathrm{~mm}$ ) or $\mathrm{CC}\left(\mathrm{SiO}_{2}, 6 \mathrm{~g} / 200 \mathrm{mg}\right.$ of crude product) with $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} /$ acetone $97: 1.5: 1.5$ as the eluent to give pure $\mathbf{3}^{\prime}$. Anal. samples were obtained by recrystallization.

6,7-Bis( benzoyloxy)-1,2-dihydro-2,9-bis(4-methoxybenzyl)-4,1-(iminomethano) isoquinoline-3,10(4H)-dione
( $\mathbf{3}^{\prime} \mathbf{a}$ ): Yield $89 \%$. Recrystallized from $\mathrm{MeOH} / \mathrm{Et}_{2} \mathrm{O}$. M.p. 144 and $146^{\circ} . \Delta H=34.4 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table ;; additionally: $8.01,7.55,7.37(2 \mathrm{PhCO}) ; 7.01,6.84\left(2 \mathrm{MeOC}_{6} \mathrm{H}_{4}\right) ; 3.78(s, 2 \mathrm{MeO})$. Anal. calc. for $\mathrm{C}_{40} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{O}_{8}$ (668.70): С 71.85, H 4.82, N 4.18; found: C 71.9, H 4.8, N 4.15.

6,7-Bis (butanoyloxy)-1,2-dihydro-2,9-bis (4-methoxybenzyl)-4,1-(iminomethano) isoquinoline $-3,10(4 \mathrm{H})$ dione ( $\mathbf{3}^{\prime} \mathbf{c}$ ): Yield $87 \%$. Recrystallized from $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{Et}_{2} \mathrm{O}$. M.p. $111^{\circ} . \Delta H=44.6$. HPLC (Chiralcel OC, $\mathbf{2}^{\prime} \mathbf{b}$ for conditions): 2 peaks, $t_{\mathrm{R}} 15$ and 18 min . ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 1; additionally: 6.95, $6.81\left(2 \mathrm{MeOC}_{6} \mathrm{H}_{4}\right) ; 3.78$ $(s, 2 \mathrm{MeO}) ; 2.48\left(t, 2 \mathrm{CH}_{2} \mathrm{CO}\right) ; 1.75\left(m 2 \mathrm{CH}_{2}\right) ; 1.02(t, 2 \mathrm{Me})$. Anal. calc. for $\mathrm{C}_{34} \mathrm{H}_{36} \mathrm{~N}_{2} \mathrm{O}_{8}(600.67): \mathrm{C} 67.99$, H 6.04, N 4.66; found: C 67.8, H 6.1, N 4.7 .

6,7-Bis(hexanoyloxy)-1,2-dihydro-2,9-bis(4-methoxybenzyl)-4,1-(iminomethano)isoquinoline-3,10(4H)-dione ( $\mathbf{3}^{\prime}$ d): Yield $84 \%$. Recrystallized from MeOH. M.p. $102-104^{\circ} . \Delta H=40 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 1 ; additionally: 6.95, $6.81\left(2 \mathrm{MeOC}_{6} \mathrm{H}_{4}\right) ; 3.79(s, 2 \mathrm{MeO}) ; 2.49\left(t, 2 \mathrm{CH}_{2} \mathrm{CO}\right) ; 1.71\left(m, 2 \mathrm{CH}_{2} \mathrm{Me}\right) ; 1.37\left(m, 4 \mathrm{CH}_{2}\right) ; 0.96$ ( $t, 2 \mathrm{Me}$ ). Anal. calc. for $\mathrm{C}_{38} \mathrm{H}_{44} \mathrm{~N}_{2} \mathrm{O}_{8}$ (656.78): C 69.49, H 6.75, N 4.27; found: C 69.5, H 6.7, N 4.3.

6,7-Bis(dodecanoyloxy)-1,2-dihydro-2,9-bis(4-methoxybenzyl)-4,1-(iminomethano) isoquinoline-3,10(4H)-dione ( $\mathbf{3}^{\prime} \mathbf{e}$ ): Yield $90 \%$. Recrystallized from $\mathrm{Et}_{2} \mathrm{O} /$ pentane. M.p. $77^{\circ}$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 1 ; additionally: 6.95, $6.81\left(2 \mathrm{MeOC}_{6} H_{4}\right) ; 3.78(s, 2 \mathrm{MeO}) ; 2.49\left(t, 2 \mathrm{CH}_{2} \mathrm{CO}\right) ; 1.70\left(m, 2 \mathrm{CH}_{2} \mathrm{Me}\right) ; 1.27\left(m, 16 \mathrm{CH}_{2}\right) ; 0.89$ ( $t, 2$ Me). Anal. calc. for $\mathrm{C}_{50} \mathrm{H}_{68} \mathrm{~N}_{2} \mathrm{O}_{8}$ (825.10): C 72.78, H 8.31, N 3.40; found: C 72.7, H 8.4, N 3.4.

6,7-Bis(benzoyloxy)-1,2-dihydro-4,1-(iminomethano)isoquinoline-3,10(4H)-dione (3a). To a stirred soln. of $\mathbf{3}^{\prime} \mathrm{a}(200 \mathrm{mg}, 0.3 \mathrm{mmol})$ in $\mathrm{MeCN}(4 \mathrm{ml})$ was added dropwise ( 30 min ) a soln. of CAN ( $1 \mathrm{~g}, 1.8 \mathrm{mmol}$ ) in $\mathrm{H}_{2} \mathrm{O}$ $(4 \mathrm{ml})$. Stirring was continued for 2.5 h , and $1 \mathrm{~N} \mathrm{NaOH}(3 \mathrm{ml})$ was added. The resulting precipitate was filtered and rinsed with MeCN and the filtrate evaporated at r.t. $\mathrm{CC}\left(\mathrm{SiO}_{2}, 25 \mathrm{~g}, \mathrm{CHCl}_{3} / \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}\right.$ 10:5:1) followed by recrystallization from $\mathrm{MeOH} / \mathrm{Et}_{2} \mathrm{O}$ gave $3 \mathrm{a}\left(56 \mathrm{mg}, 44 \%\right.$ ). M.p. $234^{\circ}$ (dec.; $10^{\circ} / \mathrm{min}$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right)$ : Table 2; additionally: 7.94, 7.62, $7.47(2 \mathrm{PhCO}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right)$ : Table 3; additionally: 163.5 (COO); 134.2, 129.5, 128.9, $127.8(\mathrm{Ph})$. MS: $429\left([M+1]^{+}\right)$. Anal. calc. for $\mathrm{C}_{24} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{6} \cdot 0.5 \mathrm{CH}_{3} \mathrm{OH}(444.42): \mathrm{C} 66.21$, H 4.08 , N 6.30 ; found: C 66.5 , H 3.8, N 6.3 .

6,7-Diacetoxy-1,2-dihydro-4,1-(iminomethano) isoquinoline-3,10(4H)-dione (3b). To a stirred soln. of $\mathbf{3} \mathbf{3} \mathbf{b}$ ( $720 \mathrm{mg}, 1.3 \mathrm{mmol}$ ) in $\mathrm{MeCN}(12 \mathrm{ml})$ was added dropwise ( 15 min ) a soln. of $\mathrm{CAN}(2.9 \mathrm{~g}, 5.3 \mathrm{mmol})$ in $\mathrm{H}_{2} \mathrm{O}$ $(12 \mathrm{ml})$. Stirring was continued for $15-30 \mathrm{~min}$. The soln. was concentrated to $\mathrm{ca} .1 / 2$ the volume at r .t., and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(10 \mathrm{ml})$ was added to give a crystalline precipitate of pure $3 \mathrm{~b}\left(230 \mathrm{mg}, 58 \%\right.$ ). M.p. $256^{\circ}$ (dec.; $10^{\circ} / \mathrm{min}$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (( $\left.\mathrm{D}_{6}\right) \mathrm{DMSO}$ ): Table 2; additionally: 2.28 ( $s, 2 \mathrm{MeCO}$ ). Anal. calc. for $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{6}$ (304.26): C 55.27, H 3.98, N 9.21 ; found: C 55.4, H 3.9, N 9.3.

6,7-Bis(butanoyloxy)-1,2-dihydro-4,1-(iminomethano) isoquinoline-3,10(4H)-dione (3c). As described for 3b, from $\mathbf{3}^{\prime} \mathbf{c}(600 \mathrm{mg}, 1 \mathrm{mmol})$. After the reaction, MeCN was evaporated at r.t. and $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{Et}_{2} \mathrm{O} /$ pentane added. The resulting precipitate was collected by filtration, washed with pentane ( 320 mg ), and discolored with activated carbon (Norit) in hot i-PrOH. Recrystallization from i-PrOH afforded pure 3c ( $150 \mathrm{mg}, 41 \%$ ). M.p. 208-211 . $\Delta H=40 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right)$ : Table 2; additionally: $2.55\left(t, 2 \mathrm{CH}_{2} \mathrm{CO}\right) ; 1.62\left(d t, 2 \mathrm{CH}_{2}\right) ; 0.95(t, 2 \mathrm{Me})$. Anal. calc. for $\mathrm{C}_{18} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{6}$ (360.37): C 59.99, H 5.59, N 7.77; found: C 59.7, H 5.6, N 7.6.

Resolution of 3c by HPLC. Prochrom column: length 23 cm and diameter 6 cm ; stationary phase: Chiralcel OC ( 500 g ); eluent: heptane $/ \mathrm{EtOH} 1: 1$; flow rate $50 \mathrm{ml} / \mathrm{min}$; UV detection at 254 nm . Three charges ( 800 mg each) of ( $\pm$ )-3c gave $(+)-3 \mathrm{c}\left(826 \mathrm{mg}, 69 \% ;[\alpha]_{\mathrm{D}}^{20}=+76 \pm 2(c=0.5, \mathrm{MeOH})\right.$, e.e. $\left.\geq 98 \%\right)$ and $(-)-3 \mathrm{c}(818 \mathrm{mg}, 68 \%$; $[\alpha]_{\mathrm{D}}^{20}=-77 \pm 2(c=0.5, \mathrm{MeOH})$, e.e. $\left.\geq 98 \%\right)$. Recrystallized from i-PrOH. M.p. $173.5^{\circ} . \Delta H=32.2$.

6,7-Bis( hexanoyloxy)-1,2-dihydro-4,1-(iminomethano) isoquinoline-3,10(4H)-dione (3d). As described for 3b, from $\mathbf{3}^{\prime} \mathbf{d}$ ( $328 \mathrm{mg}, 0.5 \mathrm{mmol}$ ). Purification by $\mathrm{CC}\left(\mathrm{SiO}_{2}, 20 \mathrm{~g}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ and then $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 80: 20\right)$ followed by TLC $\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} /\right.$ acetone $\left.80: 10: 10\right)$ gave 3d ( $124 \mathrm{mg}, 59 \%$ ). M.p. $148^{\circ}, \Delta H=36 .{ }^{1} \mathrm{H}-\mathrm{NMR}$ $\left(\mathrm{CDCl}_{3}\right)$ : Table 1 ; additionally: $7.28(m, \mathrm{NH}) ; 2.53\left(t, 2 \mathrm{CH}_{2} \mathrm{CO}\right) ; 1.72\left(m, 2 \mathrm{CH}_{2}\right) ; 1.37\left(m, 4 \mathrm{CH}_{2}\right) ; 0.93(t, 2$ Me). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right):$ Table 2; additionally: $2.56\left(t, 2 \mathrm{CH}_{2} \mathrm{CO}\right) ; 1.60\left(m, 2 \mathrm{CH}_{2}\right) ; 1.32\left(m, 4 \mathrm{CH}_{2}\right) ; 0.87$ ( $t, 2 \mathrm{Me}$ ). Anal. calc. for $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{6}$ (416.47): C $63.45, \mathrm{H} 6.78, \mathrm{~N} 6.73$; found: C $63.6, \mathrm{H} 6.9, \mathrm{~N} 6.8$.

6,7-Bis(dodecanoyloxy)-1,2-dihydro-4,1-(iminomethano)isoquinoline-3,10(4H)-dione (3e). As described for 3b, from 3'e ( $165 \mathrm{mg}, 0.2 \mathrm{mmol}$ ). Purification by 2 successive $\mathrm{TLC}\left(\mathrm{SiO}_{2}, 1 \mathrm{~mm}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} /\right.$ acetone 80:10:10) gave 3e ( $20 \mathrm{mg}, 17 \%$ ). M.p. $157.5^{\circ}\left(160.5^{\circ}, 2\right.$ nd heating). $\Delta H=50 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 1 ; additionally: 7.11 (br. $d, 2 \mathrm{NH}) ; 2.52\left(t, 2 \mathrm{CH}_{2} \mathrm{CO}\right) ; 1.71\left(m, 2 \mathrm{CH}_{2}\right) ; 1.27\left(m, 16 \mathrm{CH}_{2}\right) ; 0.88(t, 2 \mathrm{Me}) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ (( $\left.\left.\mathrm{D}_{6}\right) \mathrm{DMSO}\right):$ Table 2 ; additionally: $2.55\left(t, 2 \mathrm{CH}_{2} \mathrm{CO}\right) ; 1.56\left(m, 2 \mathrm{CH}_{2}\right) ; 1.23\left(m, 16 \mathrm{CH}_{2}\right) ; 0.84(t, 2 \mathrm{Me})$. Anal. calc. for $\mathrm{C}_{34} \mathrm{H}_{52} \mathrm{~N}_{2} \mathrm{O}_{6} \cdot 0.5 \mathrm{CH}_{3} \mathrm{OH}(600.82)$ : C 68.97, H 9.06, N 4.66; found: C 68.7, H 8.9, N 4.35.

Dimethyl (7E, 8 E )-(2,5-Dibenzyl-3,6-dioxo-2,5-diazabicyclo[2.2.2]octane-7,8-diylidene)diacetate (4'a). To a soln. of $\mathbf{2}^{\prime} \mathrm{a}(107 \mathrm{mg}, 0.25 \mathrm{mmol})$ in $\mathrm{MeCN}(4 \mathrm{ml})$ was added dropwise ( 30 min ) a soln. of CAN ( $1.64 \mathrm{~g}, 3 \mathrm{mmol}$ ) in $\mathrm{H}_{2} \mathrm{O}(4 \mathrm{ml})$ with stirring. The mixture was left at r.t. for 1 h . $\mathrm{MeCN}(10 \mathrm{ml})$ and then a soln. of $\mathrm{NaHCO}_{3}(0.84 \mathrm{~g}$, $10 \mathrm{mmol})$ in $\mathrm{H}_{2} \mathrm{O}(4 \mathrm{ml})$ were added. The resulting precipitate was filtered off and rinsed with $\mathrm{MeCN}(3 \times 10 \mathrm{ml})$ and the filtrate evaporated. The residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and the org. phase washed with $\mathrm{H}_{2} \mathrm{O}$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and evaporated: 4'a ( $48 \mathrm{mg}, 42 \%$ ). Recrystallization from $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{Et}_{2} \mathrm{O}$. M.p. $143-145^{\circ}$. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ $\left(\mathrm{CDCl}_{3}\right)$ : Table 1; additionally: 7.32, $7.12(2 \mathrm{~m}, 2 \mathrm{Ph}) ; 3.69\left(s, 2 \mathrm{CO}_{2} \mathrm{Me}\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right):$ Table 2; additionally: 7.28, $7.10(2 m, 2 \mathrm{Ph}) ; 3.57\left(s, 2 \mathrm{CO}_{2} \mathrm{Me}\right) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 3; additionally: 134.6, 128.9, 128.2, 127.9 (arom. C); $51.7(\mathrm{MeO}) ; 48.4\left(\mathrm{CH}_{2} \mathrm{~N}\right) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right):$ Table 3; additionally: 136.2, 128.4, 127.4 (arom. C); $51.5(\mathrm{MeO}) ; 47.5\left(\mathrm{CH}_{2} \mathrm{~N}\right)$. Anal. calc. for $\mathrm{C}_{26} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{6}$ (460.49): C $67.82, \mathrm{H} 5.25, \mathrm{~N} 6.08$; found: C 67.85, H 5.2, N 6.1.

Dimethyl ( $7 \mathrm{E}, 8 \mathrm{E}$ )-(3,6-Dioxo-2,5-diazabicyclo[2.2.2]octane-7,8-diylidene) diacetate (4b). To a stirred soln. of $\mathbf{2}^{\prime} \mathbf{b}(490 \mathrm{mg}, 1 \mathrm{mmol})$ in $\mathrm{MeCN}(20 \mathrm{ml})$ was added dropwise ( 30 min ) a soln. of CAN $(6.58 \mathrm{~g}, 12 \mathrm{mmol})$ in $\mathrm{H}_{2} \mathrm{O}$ ( 20 ml ). Stirring was continued for 2 h , and then a soln. of $\mathrm{Na}_{2} \mathrm{CO}_{3}(1.5 \mathrm{~g}, 14 \mathrm{mmol})$ in $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{ml})$ was added. The resulting precipitate was filtered off, rinsed with $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O} 1: 1(3 \times 25 \mathrm{ml})$, and the filtrate evaporated at r.t. $\mathrm{CC}\left(\mathrm{SiO}_{2}, 75 \mathrm{~g}, \mathrm{CHCl}_{3} / \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O} 10: 5: 1\right)$ of the residue $(9.4 \mathrm{~g})$ afforded nearly pure $\mathbf{4 b}(120 \mathrm{mg})$ in the first fractions. A second $\mathrm{CC}\left(\mathrm{SiO}_{2}, 2 \mathrm{~g}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} /\right.$ acetone $90: 5: 5$ ) gave pure $\mathbf{4 b}(95 \mathrm{mg}, 33 \%)$. Recrystallization from EtOH. M.p. $260^{\circ}$ (dec; $5^{\circ} / \mathrm{min}$ ). IR: 3240, 3020, 1730, 1700, 1430. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (( $\left.\mathrm{D}_{6}\right) \mathrm{DMSO}$ ): Table 2; additionally: $3.57\left(s, 2 \mathrm{CO}_{2} \mathrm{Me}\right) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\left(\mathrm{D}_{6}\right) \mathrm{DMSO}\right)$ : Table 3; additionally: $51.5(\mathrm{MeO})$. MS: 281
$\left([M+1]^{+}\right), 249\left([M+1-M e O H]^{+}\right)$. Anal. calc. for $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{6}(280.24)$ : C 51.43, H 4.32, N 10.00, O 34.26; found: C 51.4, H 4.3, N 9.9, O 34.2.

Dibutyl (7E,8E)-(3,6-Dioxo-2,5-diazabicyclo 2.2 .2 Joctane-7,8-diylidene)diacetate (4c). To a stirred soln. of $2^{\prime} \mathrm{c}(650 \mathrm{mg}, 1 \mathrm{mmol})$ in $\mathrm{MeCN}(10 \mathrm{ml})$ was added dropwise ( 30 min ) a soln. of CAN ( $5.5 \mathrm{~g}, 10 \mathrm{mmol}$ ) in $\mathrm{H}_{2} \mathrm{O}$ $(10 \mathrm{ml})$. Stirring was continued for 2.5 h and then the mixture extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2} . \mathrm{CC}\left(\mathrm{SiO}_{2}, 40 \mathrm{~g}, \mathrm{CHCl}_{3} /\right.$ $\left.\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O} 10: 5: 1\right)$ of the residue followed by $\mathrm{TLC}\left(\mathrm{SiO}_{2}, 2 \mathrm{~mm}, \mathrm{CHCl}_{3} / \mathrm{MeOH} /\right.$ acetone $\left.90: 5: 5\right)$ afforded pure $4 \mathrm{c}(85 \mathrm{mg}, 23 \%)$. M.p. $173^{\circ} . \Delta H 33.0 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 1 ; additionally: $4.07\left(t, 2 \mathrm{CH}_{2} \mathrm{O}\right) ; 1.63$ $\left(m, 2 \mathrm{CH}_{2} \mathrm{Me}\right) ; 1.38\left(m, 2 \mathrm{CH}_{2}\right) ; 0.93(t, 2 \mathrm{Me})$. Anal. calc. for $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{6}(364.40): \mathrm{C} 59.33, \mathrm{H} 6.64, \mathrm{~N} 7.69$; found: C 59.3, H 6.6, N 7.7.

Didodecyl (7E,8E)-(3,6-Dioxo-2,5-diazabicyclo[2.2.2]octane-7,8-diylidene)diacetate (4e). To a stirred soln. of $2^{\prime} \mathrm{f}(318 \mathrm{mg}, 0.4 \mathrm{mmol})$ in $\mathrm{MeCN} / \mathrm{THF} 1: 1(8 \mathrm{ml})$ was added dropwise ( 30 min ) a soln. of CAN ( $1.3 \mathrm{~g}, 2.4 \mathrm{mmol}$ ) in $\mathrm{H}_{2} \mathrm{O}(5 \mathrm{ml})$. Stirring was continued for 2 h , and $1 \mathrm{M} \mathrm{Na}_{2} \mathrm{CO}_{3}(2.4 \mathrm{ml})$ was added. The resulting precipitate was filtered off and rinsed with MeCN and THF and the filtrate evaporated at r.t. $\mathrm{TLC}\left(\mathrm{SiO}_{2}, 2 \mathrm{~mm}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} /\right.$ acetone 8:1:1) followed by $\mathrm{CC}\left(\mathrm{SiO}_{2}, 5 \mathrm{~g}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} /\right.$ acetone $\left.90: 5: 5\right)$ and recrystallization from $\mathrm{MeOH} /$ $\mathrm{Et}_{2} \mathrm{O}$ afforded $4 \mathrm{e}\left(30 \mathrm{mg}, 13 \%\right.$ ). M.p.: 2 transitions; $85^{\circ}, \Delta H=36.3 ; 125^{\circ}, \Delta H=5.0 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : Table 1 ; additionally: $4.06\left(t, 2 \mathrm{CH}_{2} \mathrm{O}\right) ; 1.64\left(m, 2 \mathrm{CH}_{2} \mathrm{Me}\right) ; 1.26\left(m, 18 \mathrm{CH}_{2}\right) ; 0.88(t, 2 \mathrm{Me})$. Anal. calc. for $\mathrm{C}_{34} \mathrm{H}_{56} \mathrm{~N}_{2} \mathrm{O}_{6}$ (588.83): C 69.35, H 9.59, N 4.76; found: C 69.2, H 9.6, N 4.7.

Dimethyl (7Z,8Z)-(2,5-Dibenzyl-3,6-dioxo-2,5-diazabicyclo[2.2.2]octane-7,8-diylidene)diacetate (24'). A refluxed soln. of $4^{\prime}$ a $(83 \mathrm{mg}, 0.18 \mathrm{mmol})$ and $\mathrm{I}_{2}(8 \mathrm{mg})$ in toluene $(15 \mathrm{ml})$ was irradiated for 2.5 h with a 200-W lamp. The solvent was evaporated and the residue purified by TLC ( $\mathrm{SiO}_{2}, 2 \mathrm{~mm}$, AcOEt/hexane 1:1, 2 elutions). Pure $24^{\prime}(70 \mathrm{mg}, 84 \%)$ was obtained from the lower fluorescent band. Recrystallization from MeOH. M.p. $\mathbf{1 7 5}^{\circ}$. $\Delta H=43.5 .{ }^{1} \mathrm{H}$-NMR $\left(\mathrm{CDCl}_{3}\right)$ : Table 1 ; additionally: 7.27, $7.18(2 \mathrm{~m}, 2 \mathrm{Ph}) ; 3.67\left(s, 2 \mathrm{CO}_{2} \mathrm{Me}\right) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ $\left(\mathrm{CDCl}_{3}\right)$ : Table 3; additionally: 135.2, 128.5, 128.4, $128.0(\operatorname{arom} . \mathrm{C}) ; 51.9(\mathrm{MeO}) ; 48.7\left(\mathrm{CH}_{2} \mathrm{~N}\right)$. Anal. calc. for $\mathrm{C}_{26} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{6}$ (460.49): C 67.82, H 5.25, N 6.08 ; found: C 67.7, H 5.2, N 6.1 .

## REFERENCES

[1] M.-J. Brienne, J. Gabard, M. Leclercq, J.-M. Lehn, M. Cesario, C. Pascard, M. Chevé, G. Dutruc-Rosset, Tetrahedron Lett. 1994, 35, 8157, and ref. cit. therein.
[2] J. B. Taylor, J. W. Lewis, M. Jacklin, J. Med. Chem. 1970, 13, 1226.
[3] C. de Luca, J. Chem. Soc., Perkin Trans. 2 1983, 1821.
[4] M. Vaultier, N. Knouzi, R. Carrié, Tetrahedron Lett. 1983, 24, 763.
[5] P. A. Sturm, D. W. Henry, P. E. Thompson, J. B. Zeigler, J. W. McCall, J. Med. Chem. 1974, 17, 481.
[6] P. A. Harland, P. Hodge, Synthesis 1984, 941.
[7] D. Ben-Ishai, I. Sataty, Z. Bernstein, Tetrahedron 1976, 32, 1571.
[8] D. Ben-Ishai, I. Sataty, N. Peled, R. Goldshare, Tetrahedron 1987, 43, 439.
[9] R. O. Clinton, S. C. Laskowski, J. Am. Chem. Soc. 1948, 70, 3135.
[10] W. Steglich, H.-G. Batz, Angew. Chem. Int. Ed. 1971, 10, 75.
[11] S. Raucher, P. Klein, Tetrahedron Lett. 1980, 21, 4061; F. S. Guizec, L. M. Wasmund, ibid. 1990, 31, 23.
[12] M. P. Cava, M. I. Levinson, Tetrahedron 1985, 22, 5061.
[13] T. W. Greene, P. G. M. Wuts, 'Protective Groups in Organic Synthesis', 2nd edn., John Wiley \& Sons, New York, 1991, p. 401.
[14] M. Botta, V. Summa, R. Saladino, R. Nicoletti, Synth. Commun. 1991, 21, 2181.
[15] R. M. Williams, E. Kwast, Tetrahedron Lett. 1989, 30, 451.
[16] Y.-F. Li, B. P. Malakel, E. Zbiral, Synlett 1992, 561.
[17] M. Y. Kim, J. E. Starrett Jr., S. M. Weinreb, J. Org. Chem. 1981, 46, 5383.
[18] N. G. Kundu, R. P. Hertzberg, S. J. Hannon, Tetrahedron Lett. 1980, 21, 1109.
[19] M. Yamaura, T. Suzuki, H. Hashimoto, J. Yoshimura, T. Okamoto, C. Shin, Bull. Chem. Soc. Jpn. 1985, 58, 1413; J. Yoshimura, M. Yamaura, T. Suzuki, H. Hashimoto, Chem. Lett. 1983, 1001.
[20] R. R. Scheline, Acta Chem. Scand. 1966, 20, 1182.
[21] K. Tori, Y. Yata, R. Muneyuki, Y. Takano, T. Tsuji, H. Tanida, Can. J. Chem. 1964, 42, 926.
[22] D. N. Butler, R. A. Snow, Can. J. Chem. 1972, 50, 795.
[23] L. M. Jackman, S. Sternhell, 'Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry', Pergamon Press, Oxford, 1969, p. 202.
[24] J.-M. Lehn, 'Supramolecular Chemistry', VCH, Weinheim, 1995, Sect. 9.7, pp. 190-192.
[25] M.-J. Brienne, J. Gabard, J.-M. Lehn, I. Stibor, J. Chem. Soc., Chem. Commun. 1989, 1868.
[26] C. Fouquey, J.-M. Lehn, A.-M. Levelut, Adv. Mater. 1990, 2, 254; T. Gulik-Krzywicki, C. Fouquey, J.-M. Lehn, Proc. Natl. Acad. Sci. U.S.A. 1993, 90, 163.
[27] A. Schouteeten, Y. Christidis, G. Mattioda, Bull. Soc. Chim. Fr. 1978, II, 248.
[28] G. V. Nair, Indian J. Chem. 1966, 4, 516.


[^0]:    ${ }^{1}$ ) Primes added to compound numbers 1-4 denotes the corresponding $N$-protected compounds.

[^1]:    ${ }^{2}$ ) We denote $\mathbf{A}$ the isomer which is the most easily isolated by crystallization from the reaction mixture.

[^2]:    ${ }^{\text {a }}$ ) Resonances for other protons, see Exper. Part.
    ${ }^{\text {b }} 5$.
    $\left.{ }^{\text {c }}\right) \quad J(A, B)[\mathrm{Hz}]: 14.7\left(\mathbf{2}^{\prime} \mathbf{b}\right) ; 14.8\left(\mathbf{2}^{\prime} \mathbf{g}, \mathbf{2}^{\prime} \mathbf{h}\right) ; 15.1\left(\mathbf{2}^{\prime} \mathbf{a}\right) ; 15.3\left(\mathbf{4}^{\prime} \mathbf{a}\right)$.
    $\left.{ }^{\text {d }}\right) m, A A^{\prime} X X^{\prime}$ spin system.

